

Partial Pressure Gauges

Quadrupole Mass Spectrometer were invented 1958 at the
University of Bonn by

Paul, Steinwedel, von Zahn

1962 Start of development in Liechtenstein for industrial use
(based on a licence from Siemens)

First applications were residual gas analysis vacuum diagnostics
because these instruments are rather small and easy to be
transported/installed compared to magnetic MS

Now these instruments are used as „intelligent sensor“ in various
industrial and scientific applications



Partial Pressure Gauges

Now „Q-poles“ are used in various applications:

- Single units for vacuum-diagnostics
- Process Monitors
- Detector for SIMS
- Bench-Top Instruments
- GC-MS for trace analysis
- Plasma Diagnostics
- Endpoint Detection for Etching in the gas phase
- Breath gas analysis
- .
- .
- Integration into environmental analysis systems



***Quadrupole Mass Spectrometer
as RGA's
in UHV and XHV***

***What is the difference compared to a partial
pressure gauge for HV-Applications?***

Partial Pressure Gauges

Scope of this lesson:

- Principle of Operation, functional units of an “RGA”
- Interpretation of Spectra
- Calibration, detectors, electron energy
- Key Features of quadrupole mass spectrometers

- → XHV/UHV

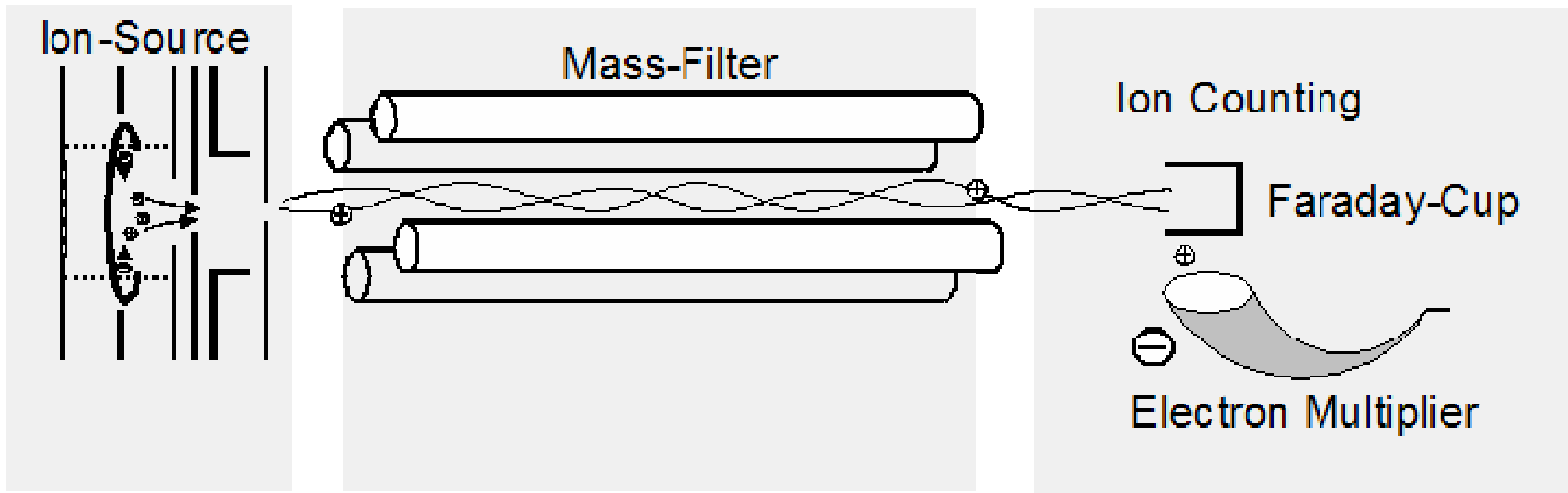
- Demands to a Partial Pressure Gauge in the UHV/XHV
- Technical Solution
- How to bake an RGA, cold spots
- Residual gas spectra in the UHV and artefacts (EID-Ions)

Partial Pressure Gauges



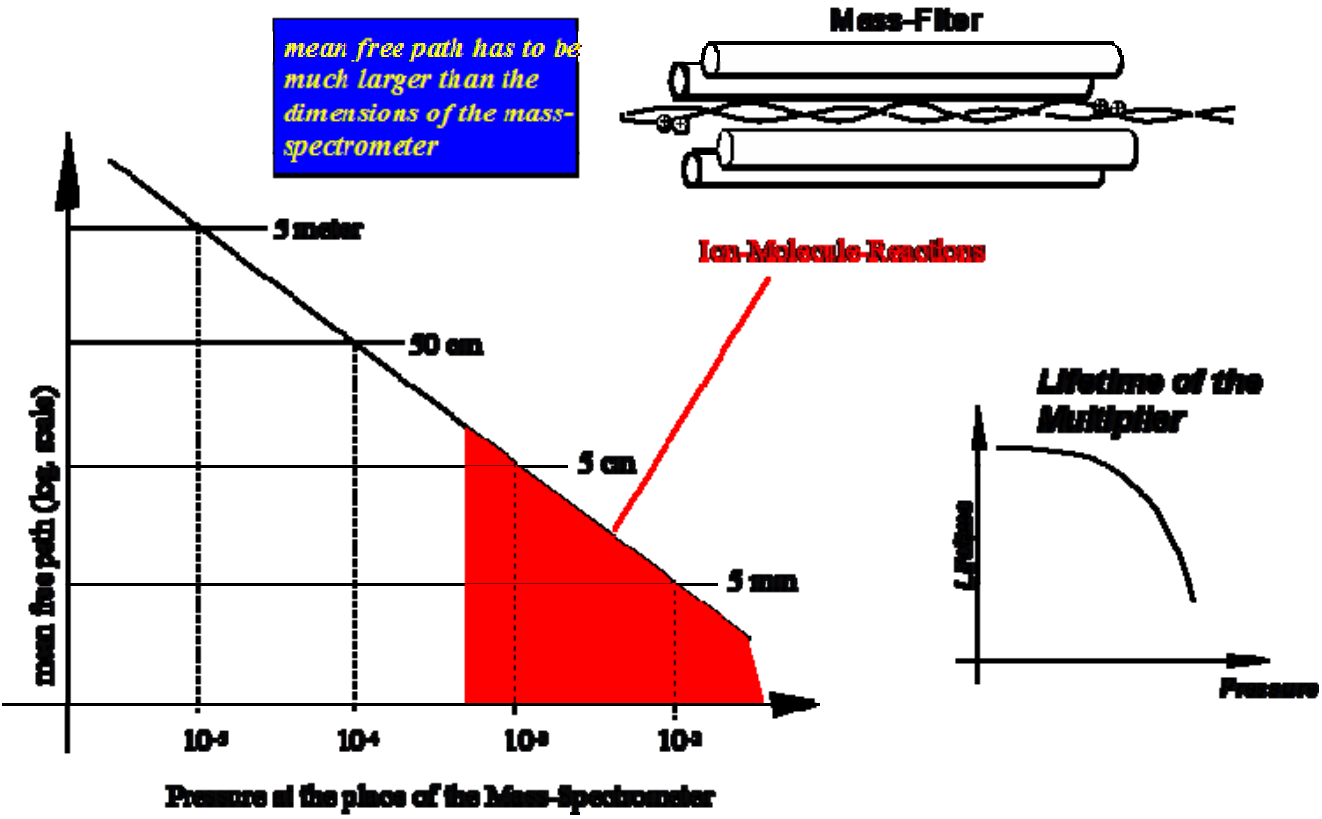
Partial Pressure Gauges

Functional units of a quadrupole mass spectrometer



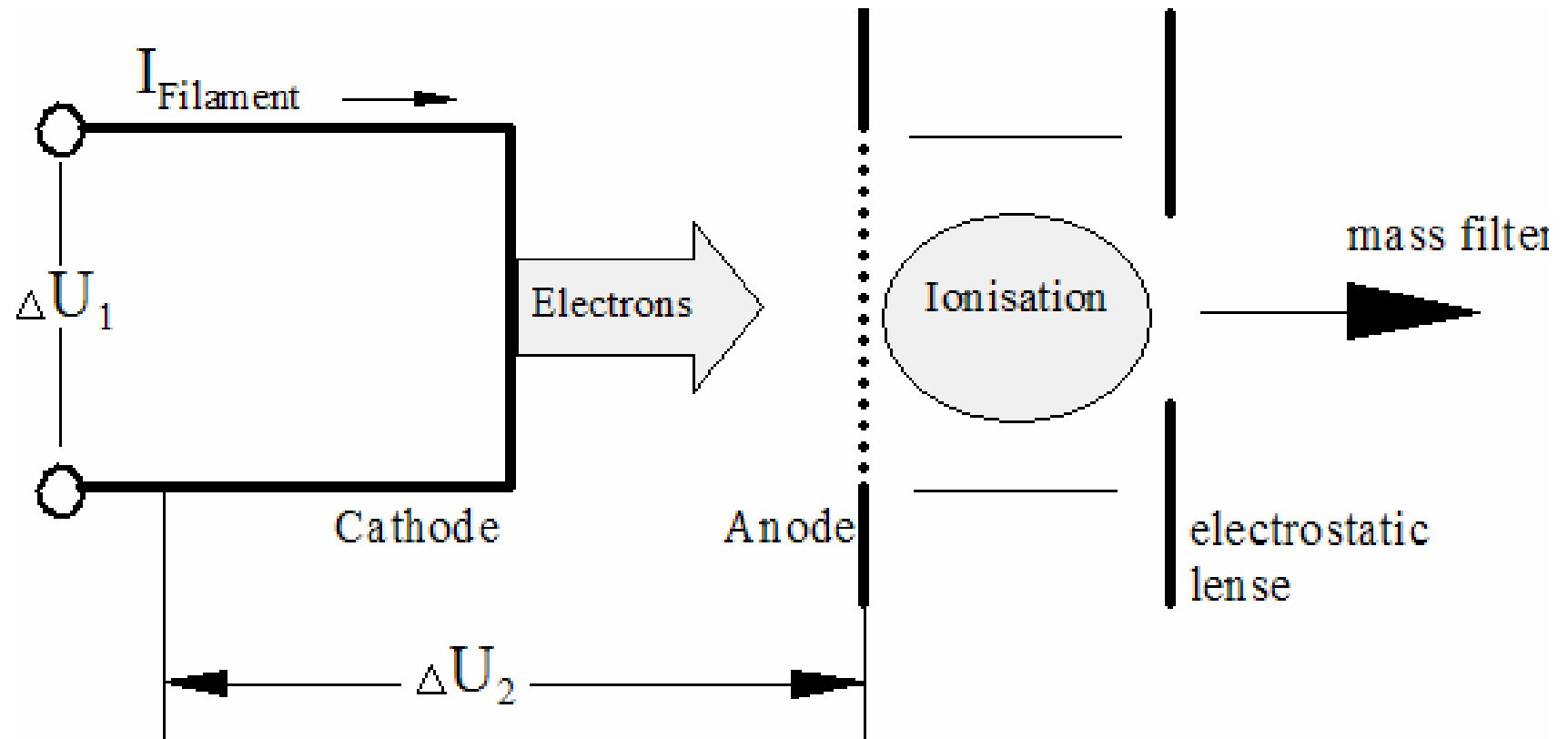
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Requirements to operate a quadrupole mass spectrometer



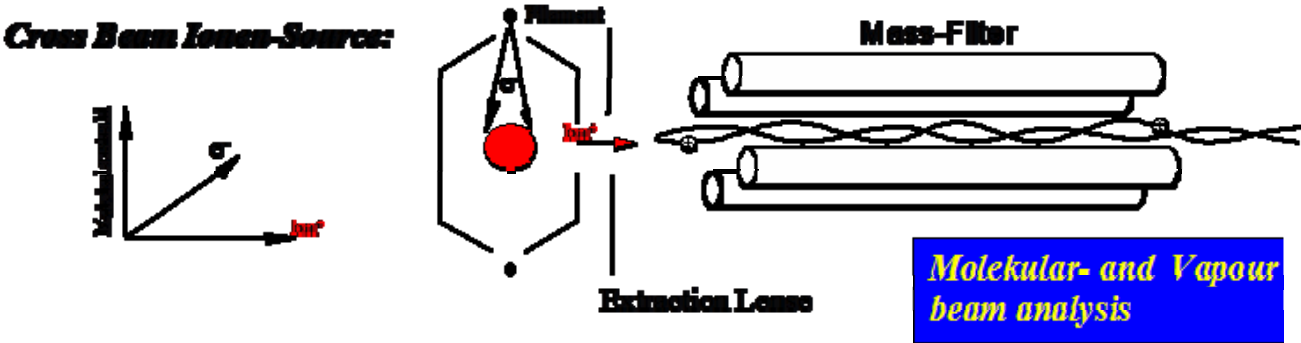
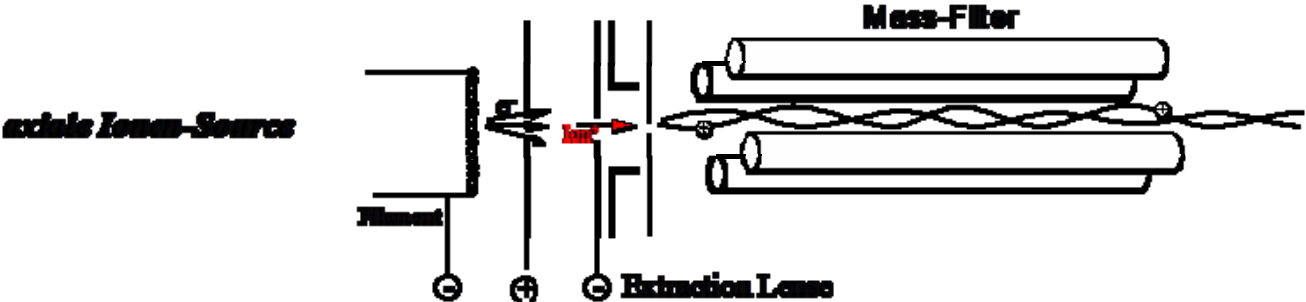
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Ionisation, Ion Sources:



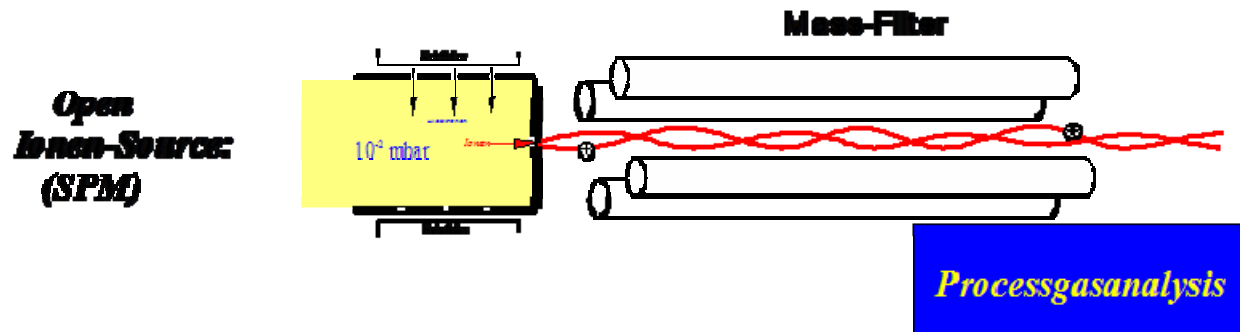
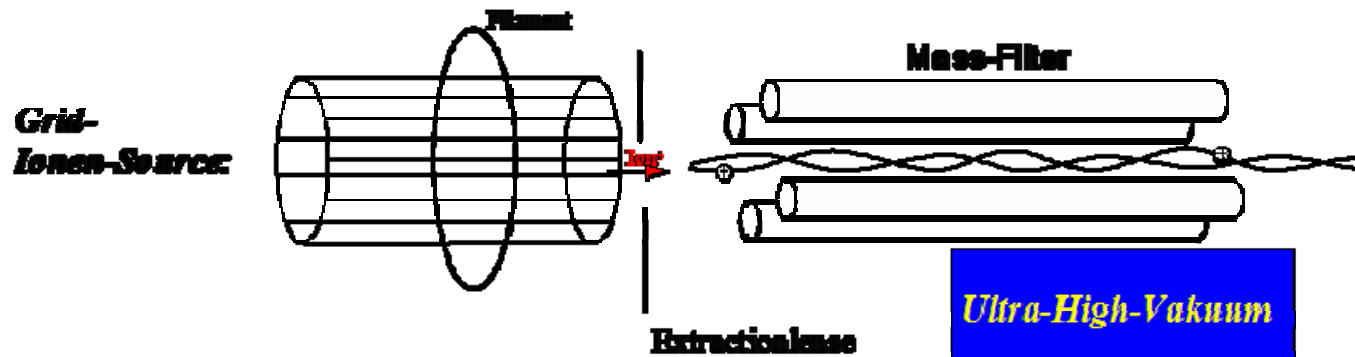
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Ionisation, Ion Sources:



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Ionisation, Ion Sources:



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Ionisation, Ion Sources:



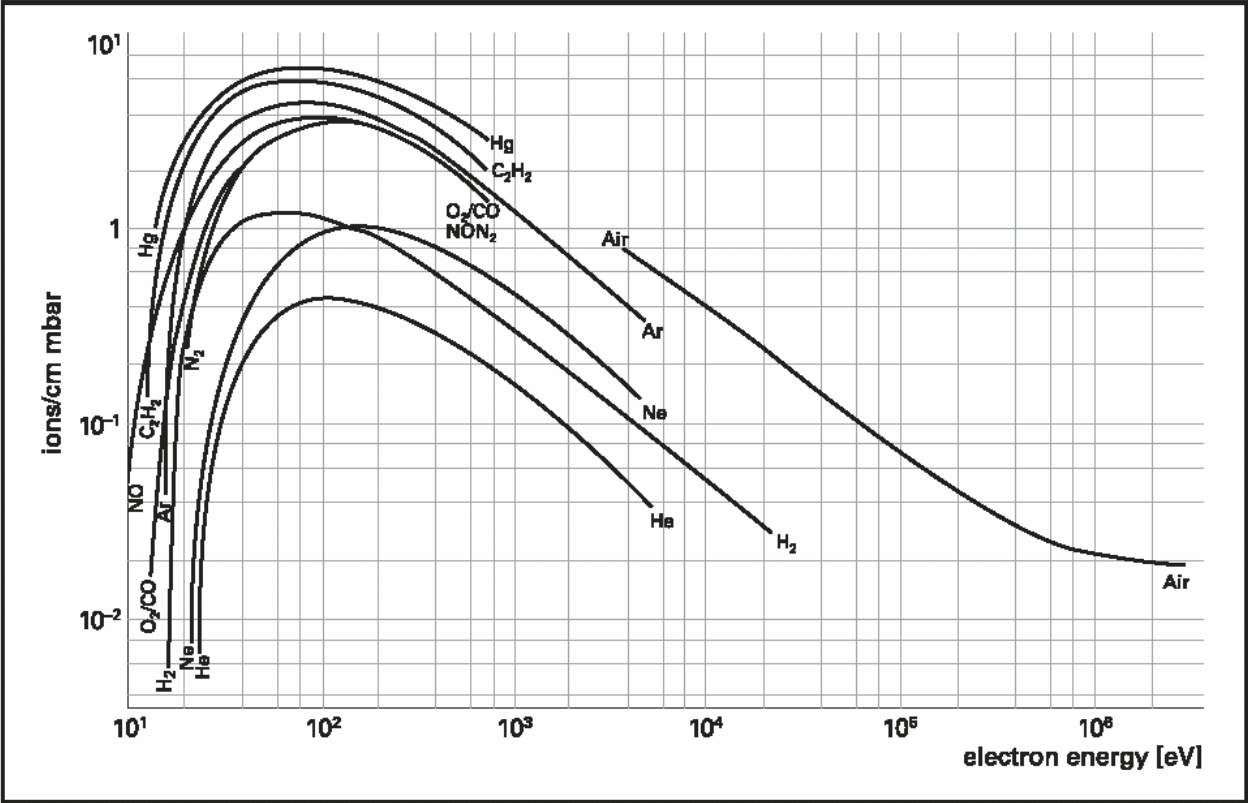
Surface is minimized,
therefore
minimum degassing
minimum artefacts from
electron impact desorption

only metal or Al_2O_3 used
gaps avoided
and
threads have additional holes
for easy degassing

Partial Pressure Gauges

Ionisation, Ion Sources:

Because electron impact ionisation is applied, it's a gas specific gauge:



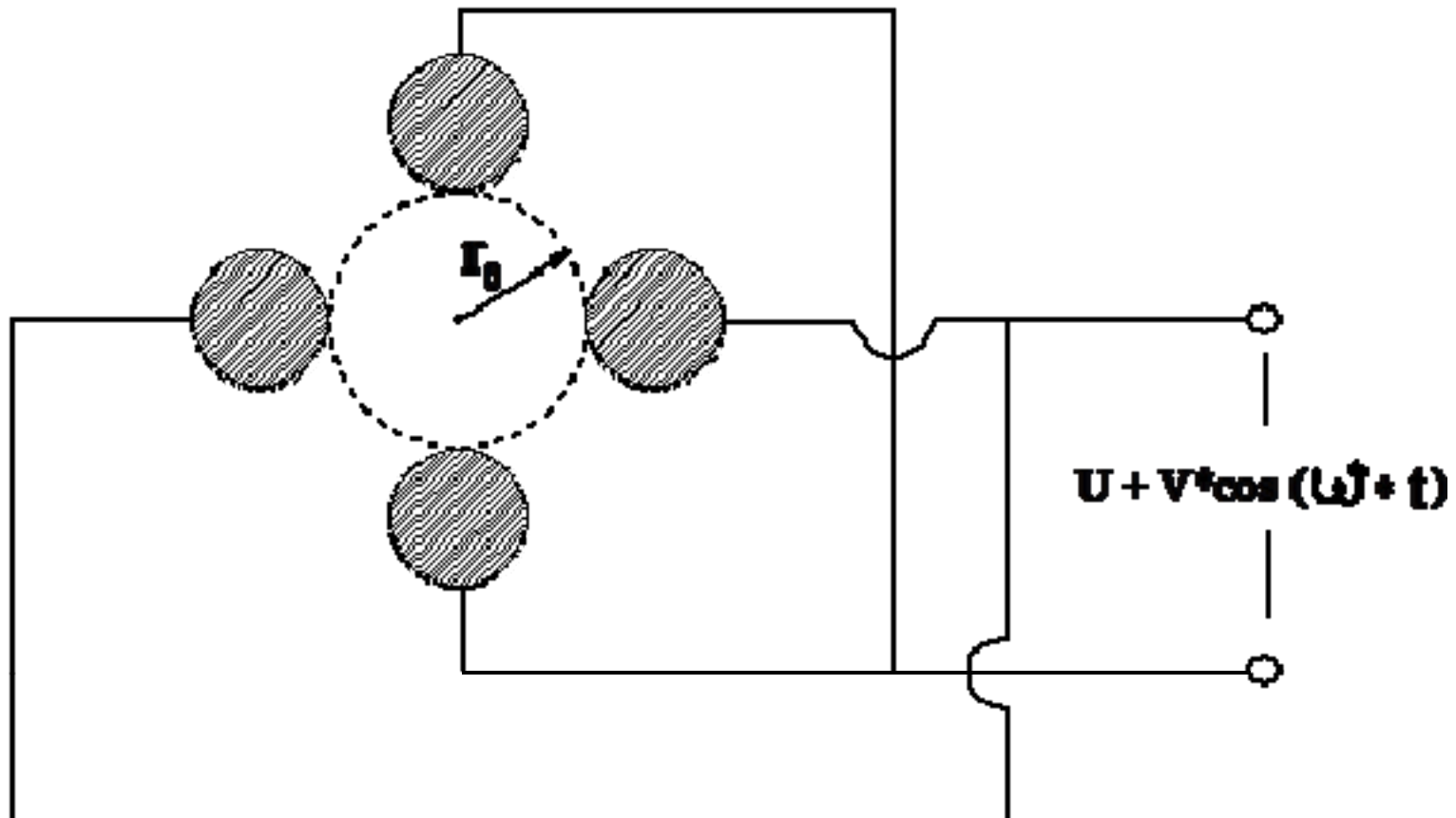
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Mass separation in an electrical field:



Partial Pressure Gauges

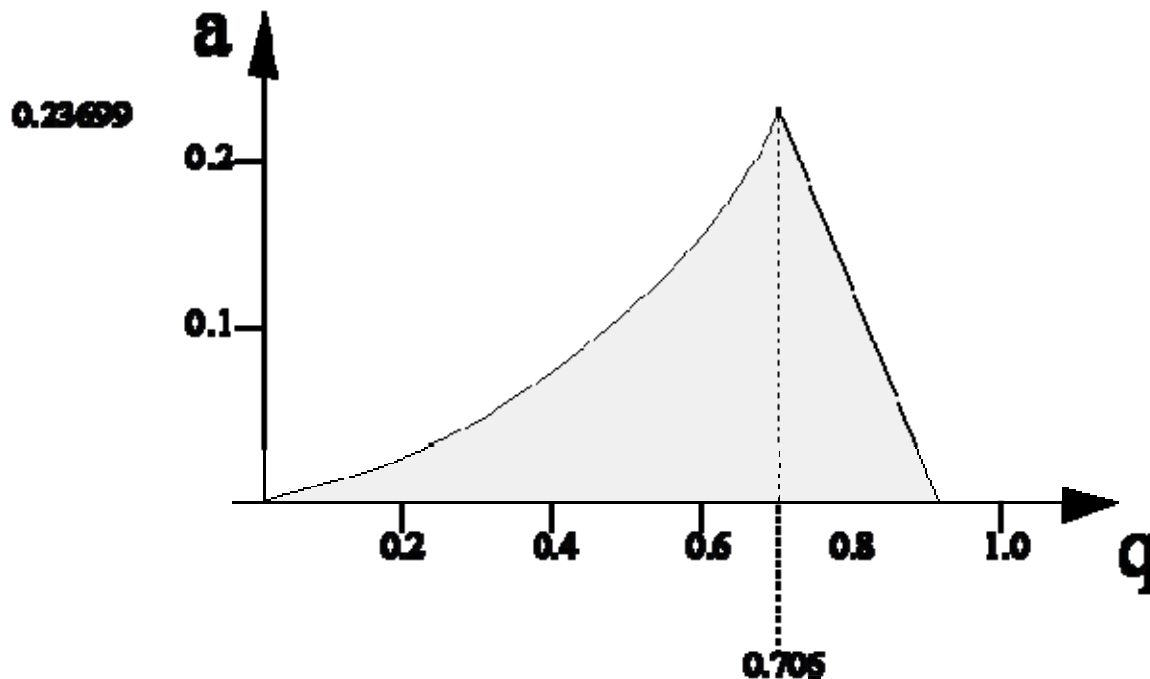
Mass separation in an electrical field:



Partial Pressure Gauges

Mass separation in an electrical field:

Only ions with a well defined m/e ratio can achieve stable trajectories in the Rf-field

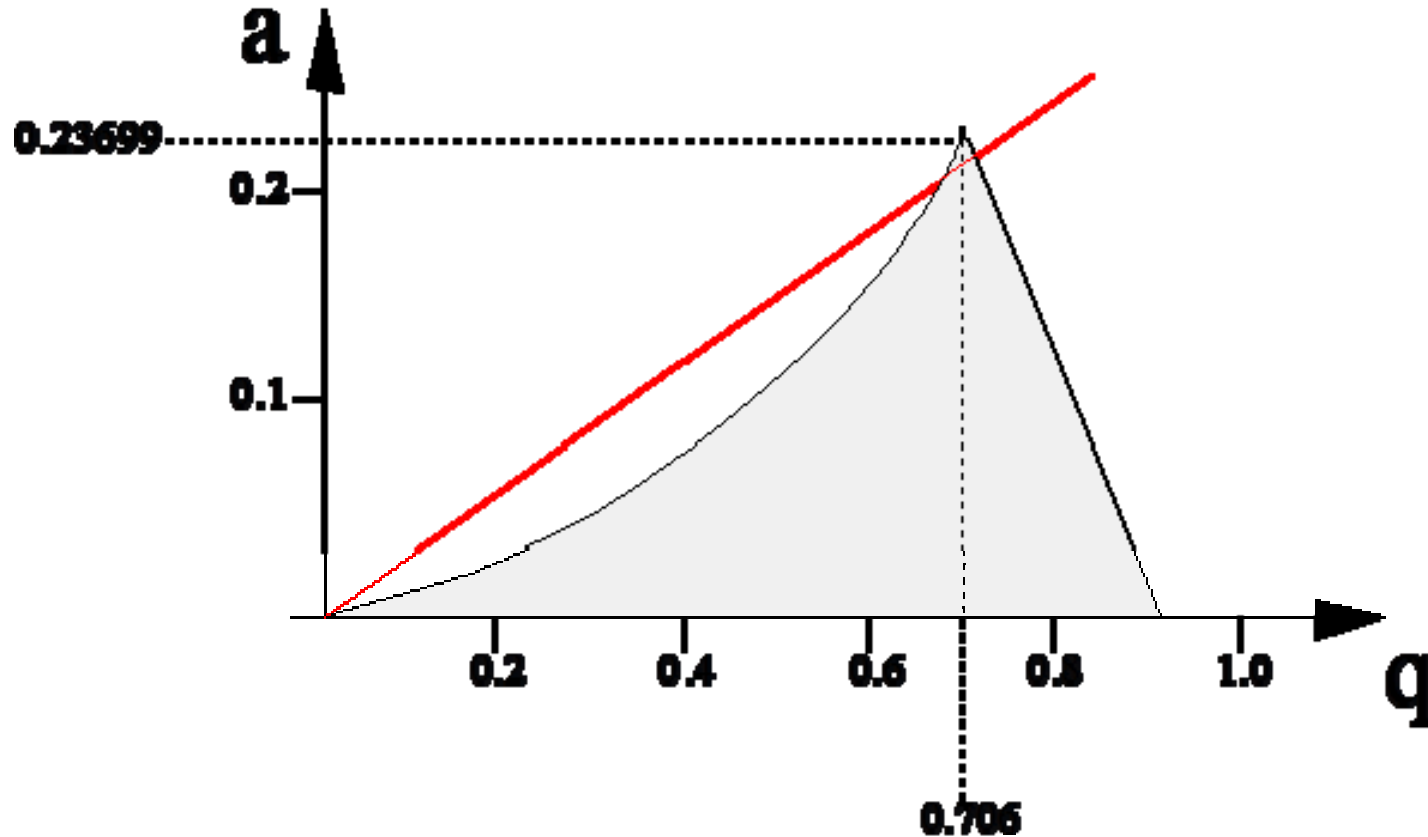


$$a = \frac{8 * e * U}{m * r_0^2 * \omega^2}$$

$$q = \frac{4 * e * V}{m * r_0^2 * \omega^2}$$

Partial Pressure Gauges

Mass separation in an electrical field:



Partial Pressure Gauges

Mass separation in an electrical field:

The ratio U/V is kept constant; the absolute value is increased starting from zero:

First ions with $m/e = 1$ enter the stability diagram, then ions with $m/e = 2$ enter the stability diagram

.....

The value of U/V which results in a stable trajectory is a linear function of the mass.

Increasing U/V from zero results in a linear mass scale

Partial Pressure Gauges

Mass separation in an electrical field:

For the user of a quadrupole mass spectrometer it is less important to be “able to solve the Mathieu’s differential equations”

Important to keep in mind:

- The required mechanical precision of the filter is in the range of microns
- The stability and precision of the applied Rf-frequency determines the performance of the mass filter in addition
- The inner range of the filter has to be absolutely clean; small dielectric particles and or coatings may lead to a poor performance

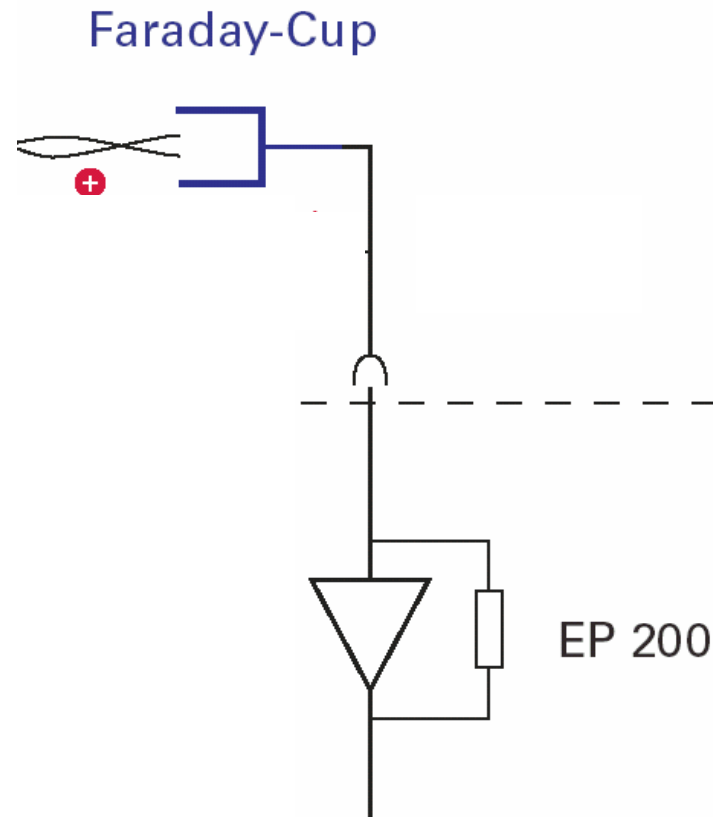
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Detectors

Ions leaving the quadrupole mass spectrometer fly into a *Faraday-Cup* and give off their charge

Charge is measured as a current by an electrometer amplifier

Measurable currents are from $1\text{E-}15$ to $1\text{E-}8$ A



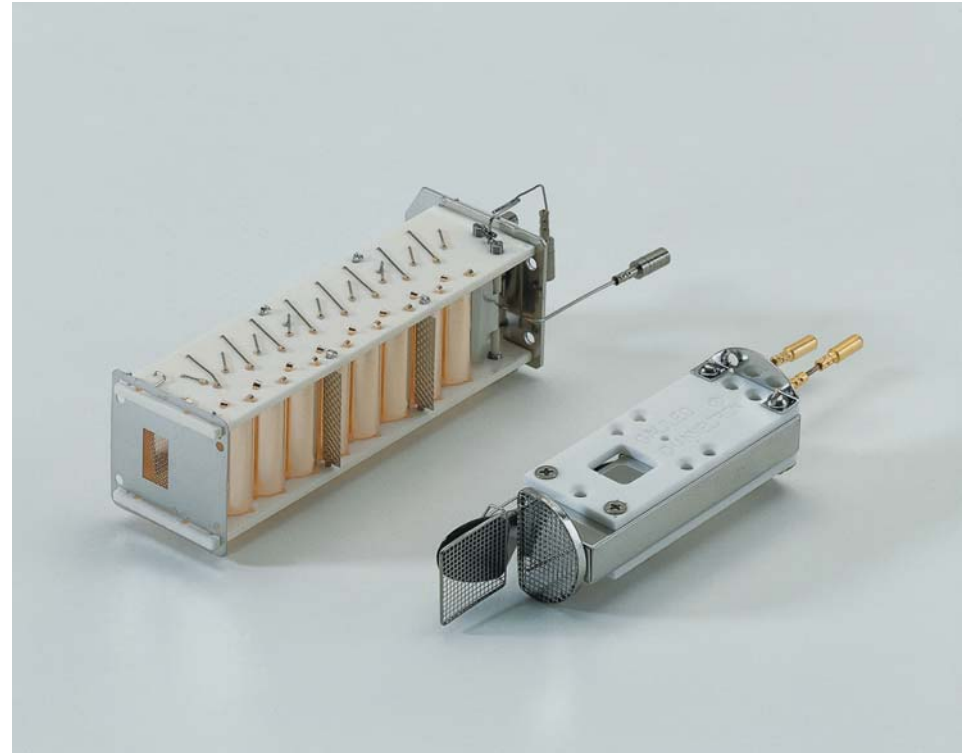
Partial Pressure Gauges

Detectors

Secondary Electron Multipliers

SEM 217,
17 discrete dynodes

Channeltron™,
continuous dynode
(incl. Faraday Detector)



Partial Pressure Gauges

Detectors

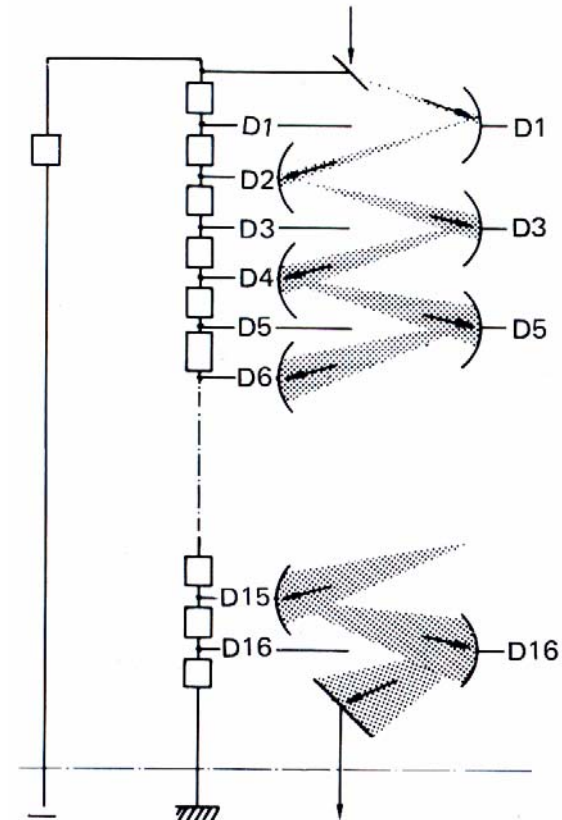
Particles (ions, neutrals, electrons, photons) hitting a surface with high energy release several „secondary electrons“

Use of several dynodes allows for an amplification of up to 10^8

Low ion currents can be detected easily

Measuring range $1\text{E-}15$ to $1\text{E-}5\text{A}$

„Counting“ of individual electron bursts allows for the detection of single ions ($1\text{E-}19\text{A}$)



Partial Pressure Gauges

Interpretation of Residual Gas Spectra

Different molecules have the same mass, for example N_2 and CO

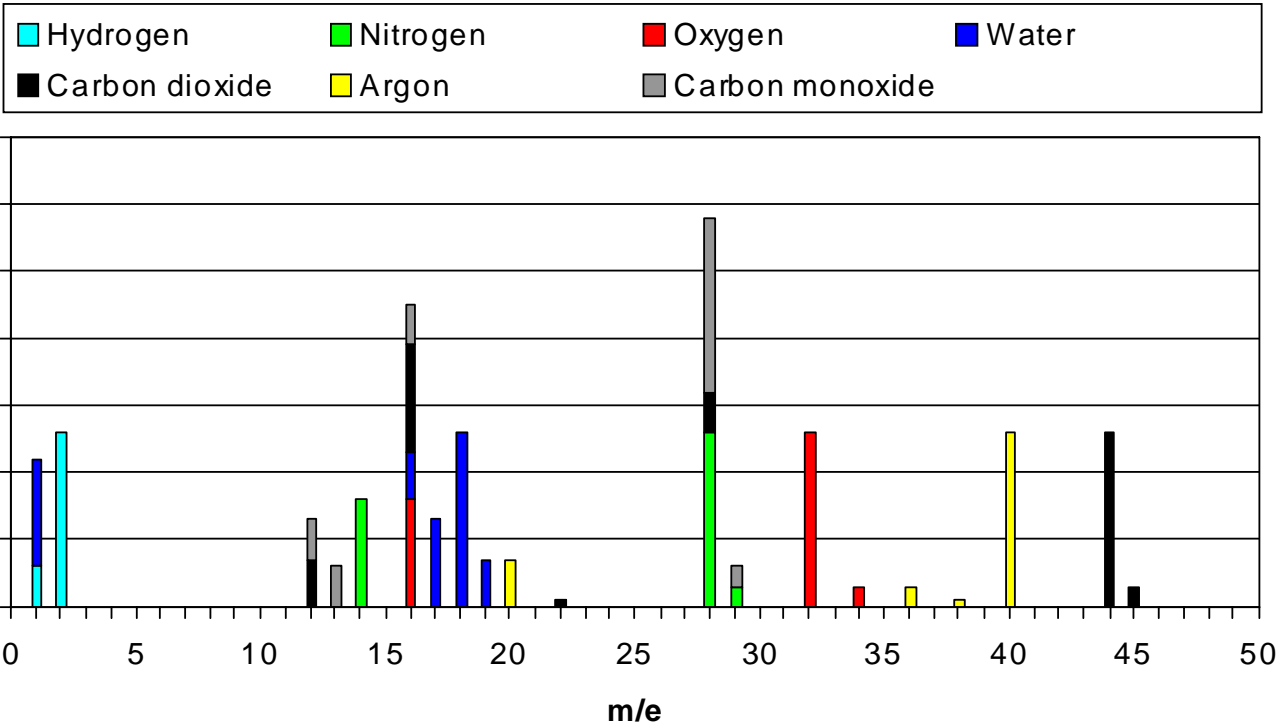
The gas molecules will be ionised and fractionised, for example water will give a signal at $m/e = 1, 2, 16, 17$ and 18 corresponding to ions such as H^+ , H_2^+ , O^+ , OH^+ and H_2O^+

This can result in rather complex spectra.

Partial Pressure Gauges

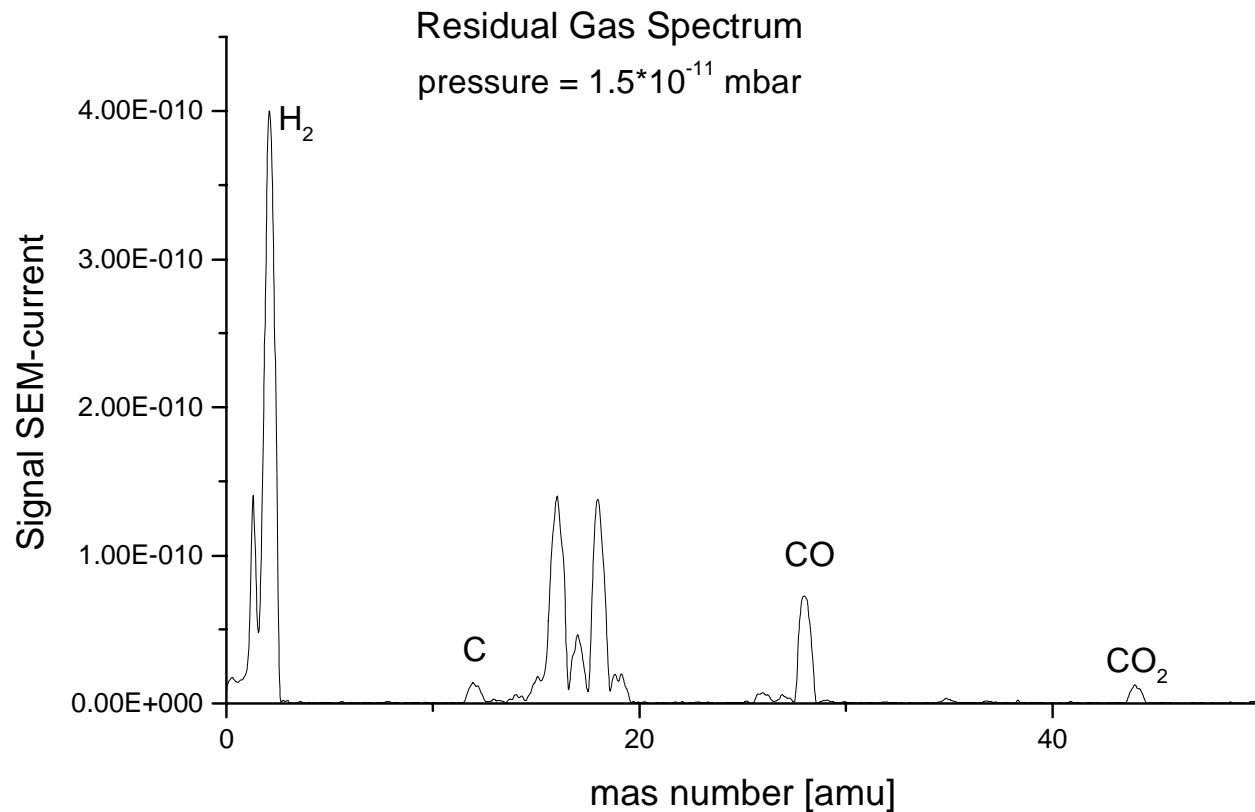
Interpretation of Residual Gas Spectra

Model spectrum (Origin of peaks)



Partial Pressure Gauges

Interpretation of Residual Gas Spectra in UHV/XHV



Partial Pressure Gauges

Interpretation of Residual Gas Spectra in UHV/XHV

The main gases are hydrogen, carbon monoxide, carbon dioxide, sometimes there are very small quantities of water vapour even in the UHV.

All other gas components can be regarded as impurities, artefacts or will indicate a leak (N₂)

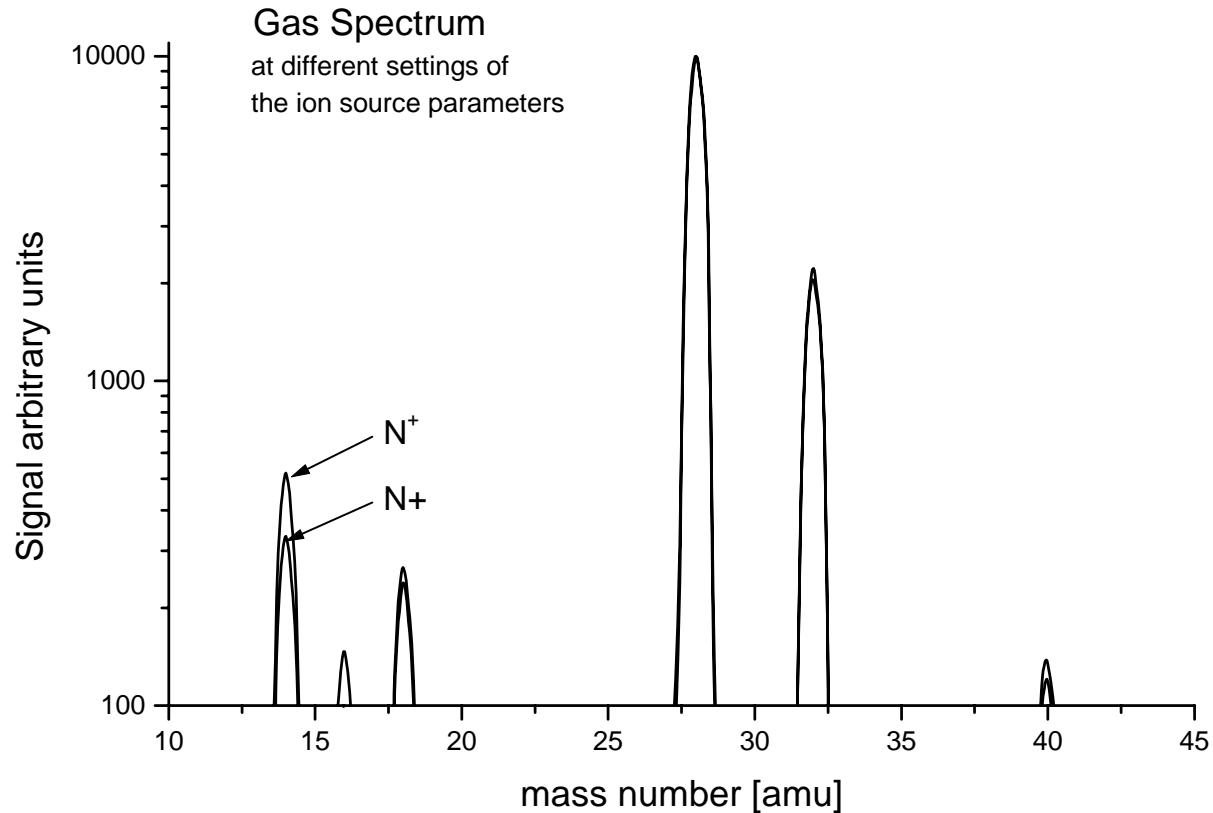
Sometimes UHV can be easier than HV!...

(It's not required to go into detail further with interferences and the like.)



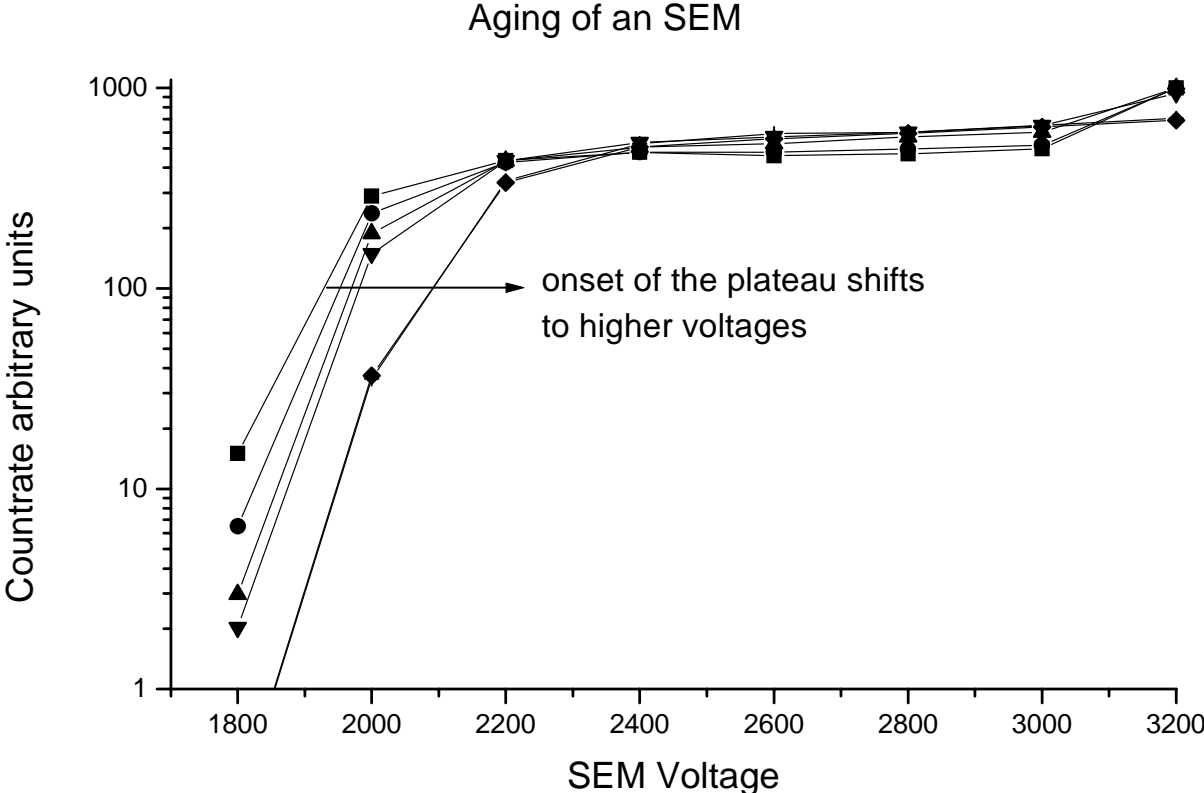
Partial Pressure Gauges

Important to know: The Influence of the Electron Energy



Partial Pressure Gauges

Important to know: The Aging of an SEM

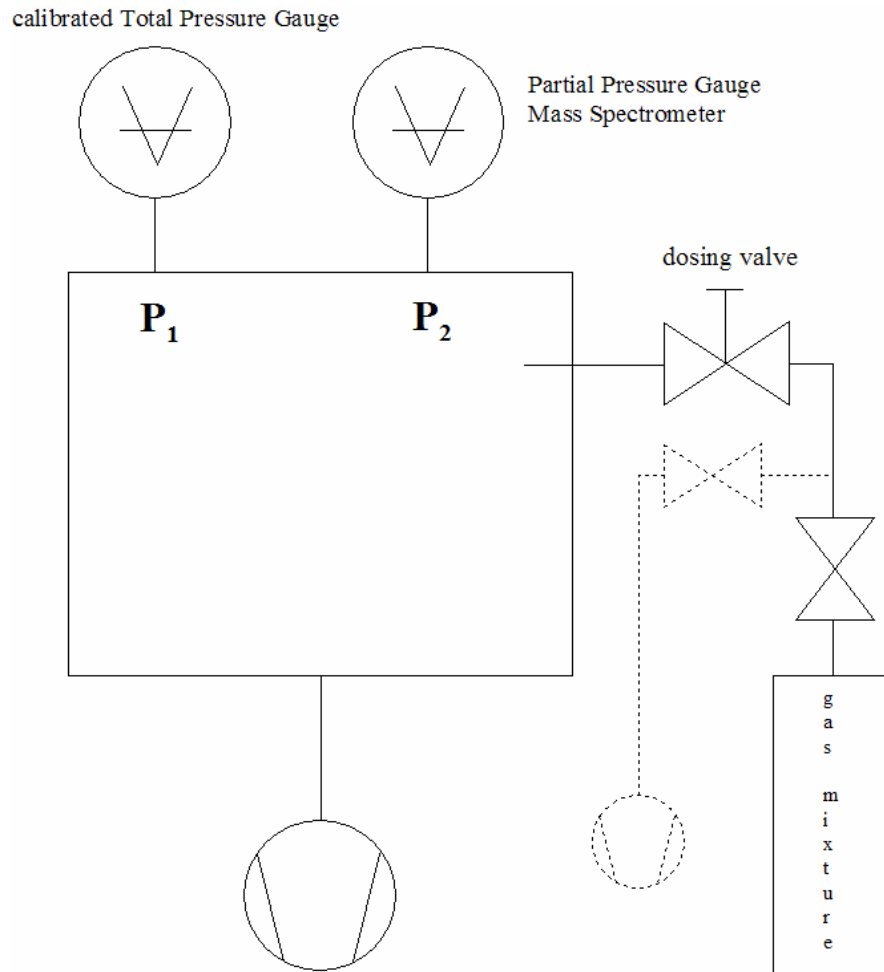


Partial Pressure Gauges

- The quadrupole mass spectrometer is gas specific.
 - The spectra obtained depend on the setting of the instrument – for example the electron energy has an influence on the spectra obtained.
 - Obviously as with every spectrometer the resolution has an influence on the signal.
 - An SEM is required for sensitivity reasons and every SEM undergoes aging.
- → Calibration is a MUST

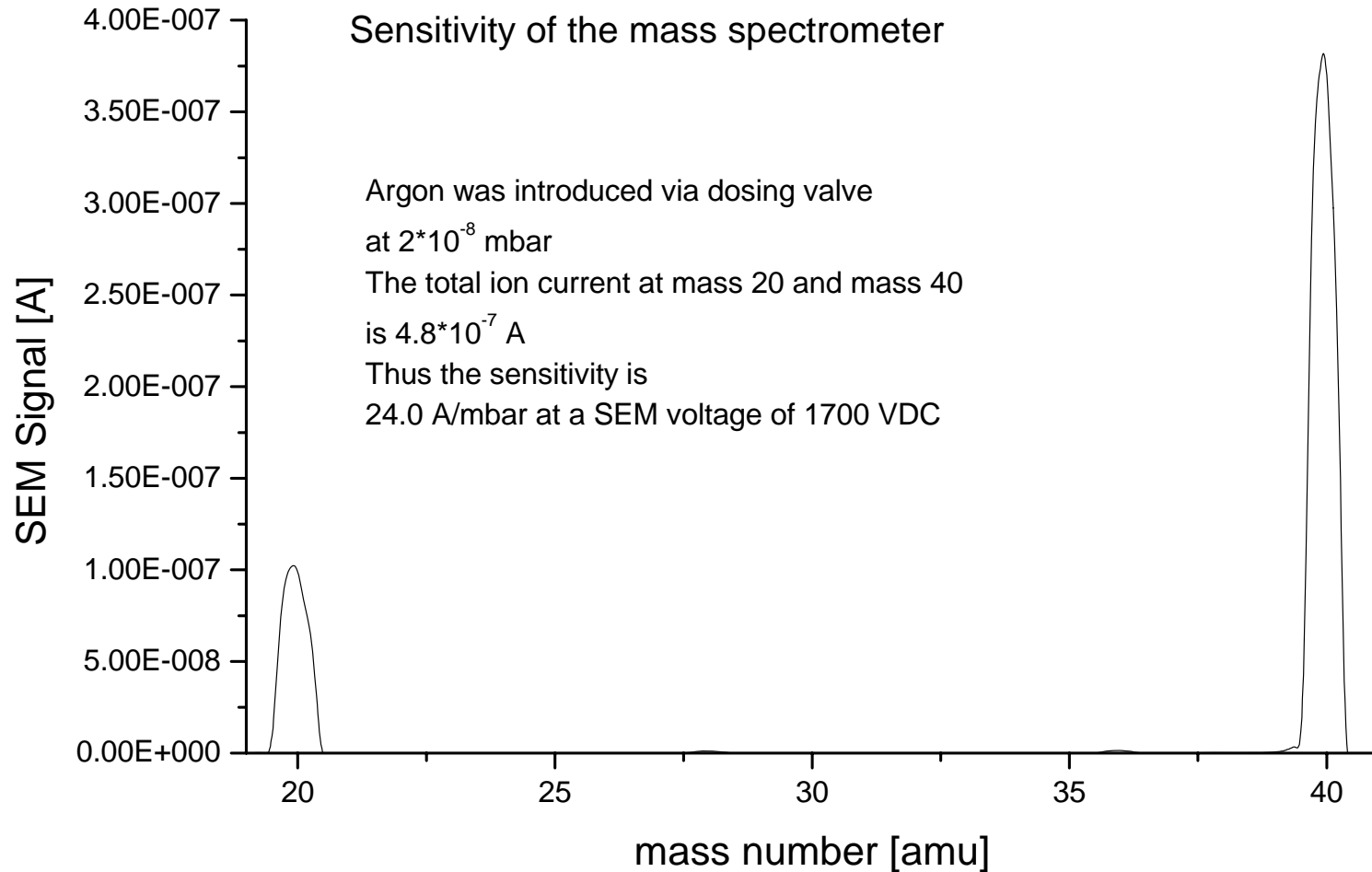
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Calibration of Mass Spectrometers



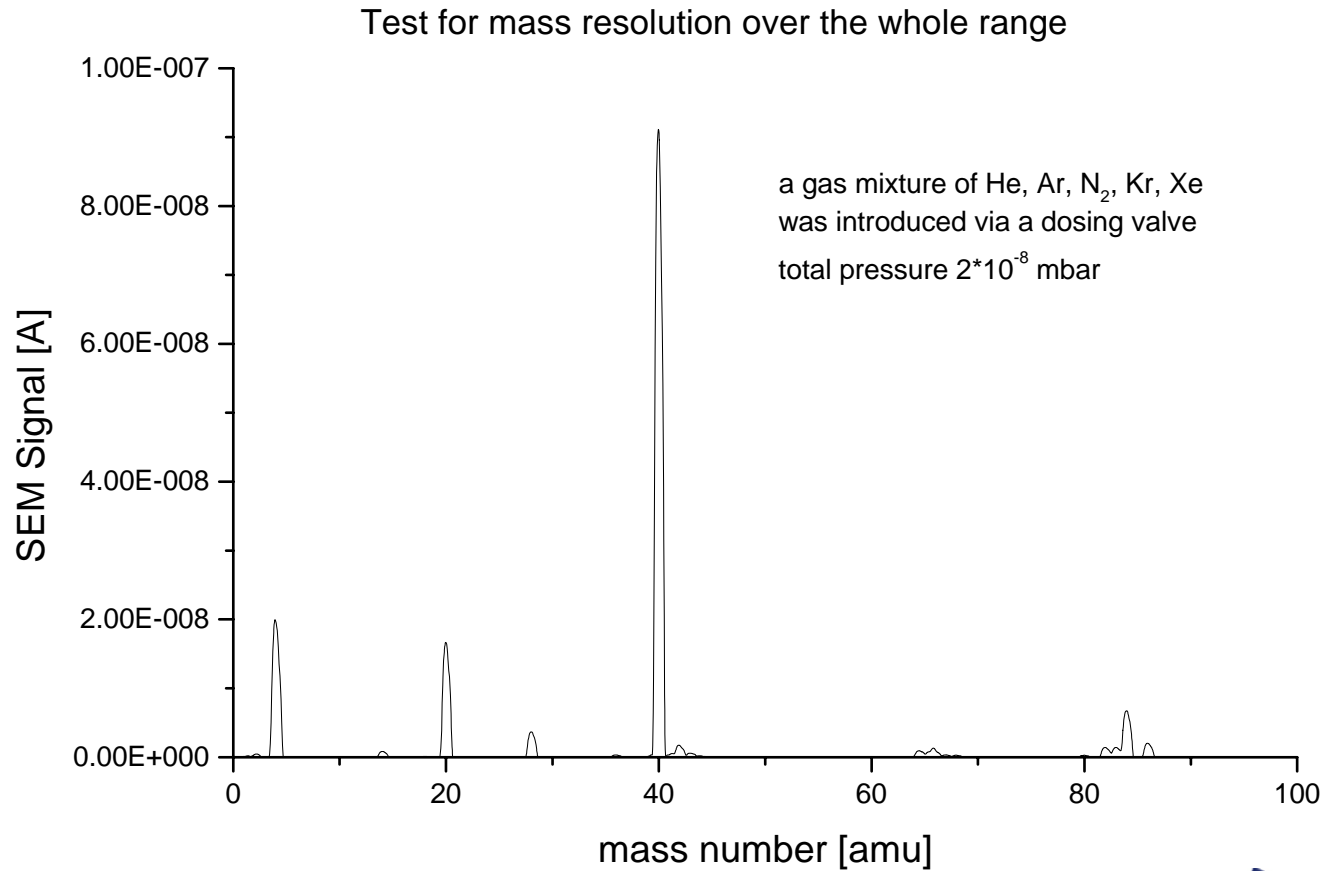
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Calibration of Mass Spectrometers



Partial Pressure Gauges

Calibration of Mass Spectrometers



Partial Pressure Gauges

Large Hadron Collider

Test Protocol demanded by
CERN

Test Report Residual Gas Analyzers

System No.5

Name:	Articlenumber:	Serialnumber:
QMA 125	PTM10777-1	44197127
QME 125 -1	PTM36376	44197979
QMS 422	PTM26580	44198861
EP 422	BG444570-T	735
Cable length QME analyzer:		0.17m

IS-Voltage QME 125	
Ion. Ref	100V
Field axis	12V
Deflection	270V
Emission	2.00mA

Measurements after a bakeout at 300° C at an UHV-chamber

- Filename:
- Leak-Test** with Helium after bakeout at 300° C
[rga nr.5 leaktest sem.bmp](#)
 - Spectrum** at base pressure with Faraday Cup, 10s/amu, 0-50amu
[rga nr.5 restgasspektrum fc.sac](#)
 - Spectrum** at base pressure with SEM, (1600 VDC) 2s/amu, 0-100amu
[rga nr.5 restgasspektrum sem.sac](#)
 - Spectrum** with Testgas (He, N2, Ar, Kr) Faraday to demonstrate the mass resolution
[rga nr.5 gasspektrum fc.sac](#)
 - Spectrum** with Testgas (He, N2, Ar, Kr) SEM 1600 VDC to demonstrate the mass resolution
[rga nr.5 gasspektrum sem.sac](#)
 - Sensitivity** for Argon Detector: Faraday 2×10^{-8} mbar argon 2s/amu
Sensitivity: **1.68E-04** A/mbar
[rga nr.5 empfindlichkeit fc.sac](#)
 - Sensitivity** for Argon, SEM at 1600 VDC 2×10^{-8} mbar argon 2s/amu
Sensitivity: **8.71** A/mbar
[rga nr.5 empfindlichkeit sem.sac](#)



Partial Pressure Gauges

Test Protocol demanded by
CERN

Detection Limit:

A signal two times the noise band of the detector determines the detection limit. For measurements with a SEM the noise band and the sensitivity are a function of the applied voltage. Therefore the sensitivity and the noise band have to be determined at various voltages to find the optimum SEM-Voltage. The noise band is determined by a measurement at mass 5.5 with a sampling time of 60 seconds for ten minutes.

8. Detection Limit Faraday	Noise band:	7.58E-15 A	rga nr.5 rauschen fc.mdc
	detection limit with Faraday Cup:	4.50E-11 mbar	

Detection Limit with SEM at various SEM-Voltages 2×10^{-8} mbar argon, gauge calibrated for N2

SEM-Voltage:

1600 VDC:	Noise band:	7.58E-14 A	Filename:	rga nr.5 rauschen 1600v.mdc
	Sensitivity:	8.71 A/mbar	Filename:	rga nr.5 empfindlichkeit 1600v.sac
	Detection Limit:	8.70E-15 mbar		
1700 VDC:	Noise band:	1.26E-13 A	Filename:	rga nr.5 rauschen 1700v.mdc
	Sensitivity:	24.49 A/mbar	Filename:	rga nr.5 empfindlichkeit 1700v.sac
	Detection Limit:	5.14E-15 mbar		
1750 VDC:	Noise band:	1.05E-13 A	Filename:	rga nr.5 rauschen 1750v.mdc
	Sensitivity:	38.66 A/mbar	Filename:	rga nr.5 empfindlichkeit 1750v.sac
	Detection Limit:	2.72E-15 mbar		



Question(s):

Which features describe such an instrument in a basic way?

Which features given in documents result from basic features or just contribute to them.

Are there data related to the whole system not only the RGA?

When data are reported, how can they be checked.

“Definitions to be complete if possible”

Partial Pressure Gauges

Key Features of Quadrupole Mass Spectrometers

- Sensitivity is gas-specific
- Mass-Resolution depends on mass range,..
- Measurement Speed depends on detection limit
- Detection Limit speed, vacuum system,..
- Accuracy standard to compare with
- Stability and Reproducibility standard to compare with



The key features of a mass spectrometer are:

- * **Detection limit** (depends on the setting of the instrument and in some cases on the vacuum system too)
- * **Dynamic Range** of the analysis
- * **Mass Resolution**, the contribution to the neighbouring mass is an appropriate definition
- * The measurement **speed**.

All these can not be discussed as single parameter, they mutual influence each others. For example the specified detection limit may never be achieved at the maximum measurement speed. A sensitivity can never be specified without the resolution of the instrument.....

Partial Pressure Gauges



Partial Pressure Gauges

So far all statements are valid for every quadrupole mass spectrometer.

Which restrictions to these instruments appear because of UHV/XHV environment?

Partial Pressure Gauges

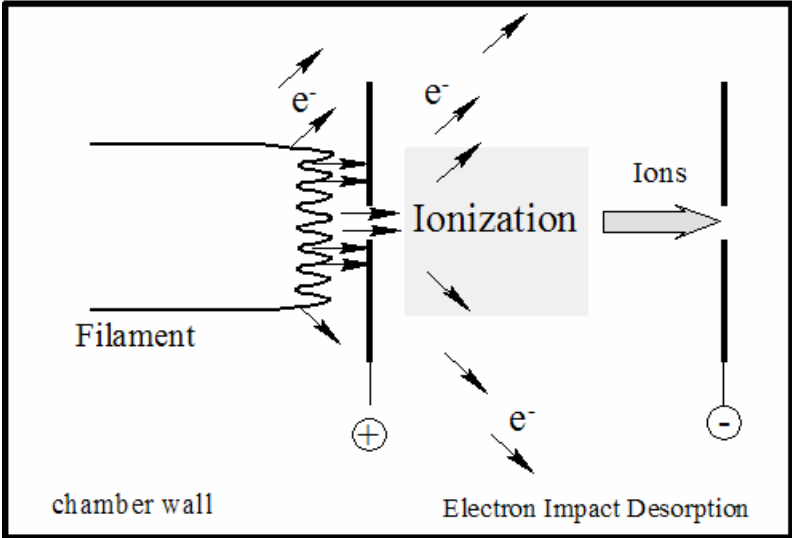
Requirements to a Partial Pressure Gauge used in UHV/XHV

- **Detection Limit $5 \cdot 10^{-15}$ mbar** ,impurities in the residual gas have to be detected, $5 \cdot 10^{-14}$ mbar is already a compromise for other reasons
- Gauge has to “survive” the baking of the vacuum system
- No influence of the gauge to the vacuum
- Gauge has to “survive radiation”

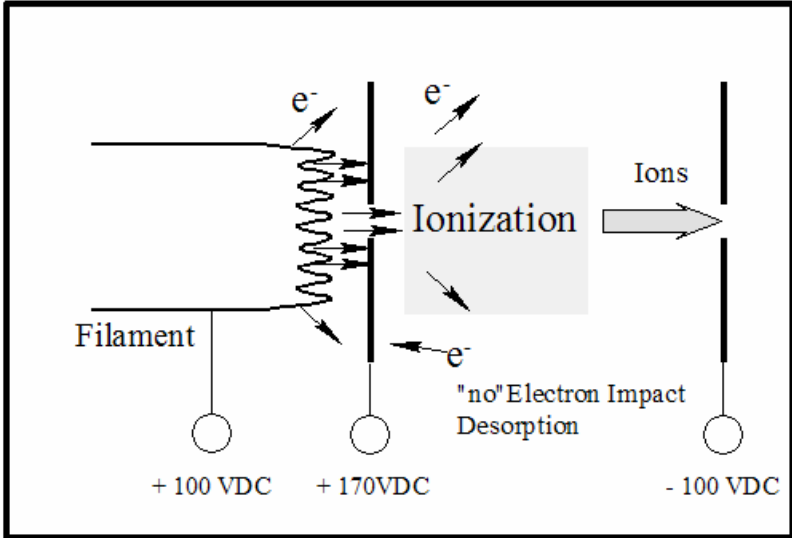
Partial Pressure Gauges

Detection Limit: Reduce Background

most used technique: no electrical bias

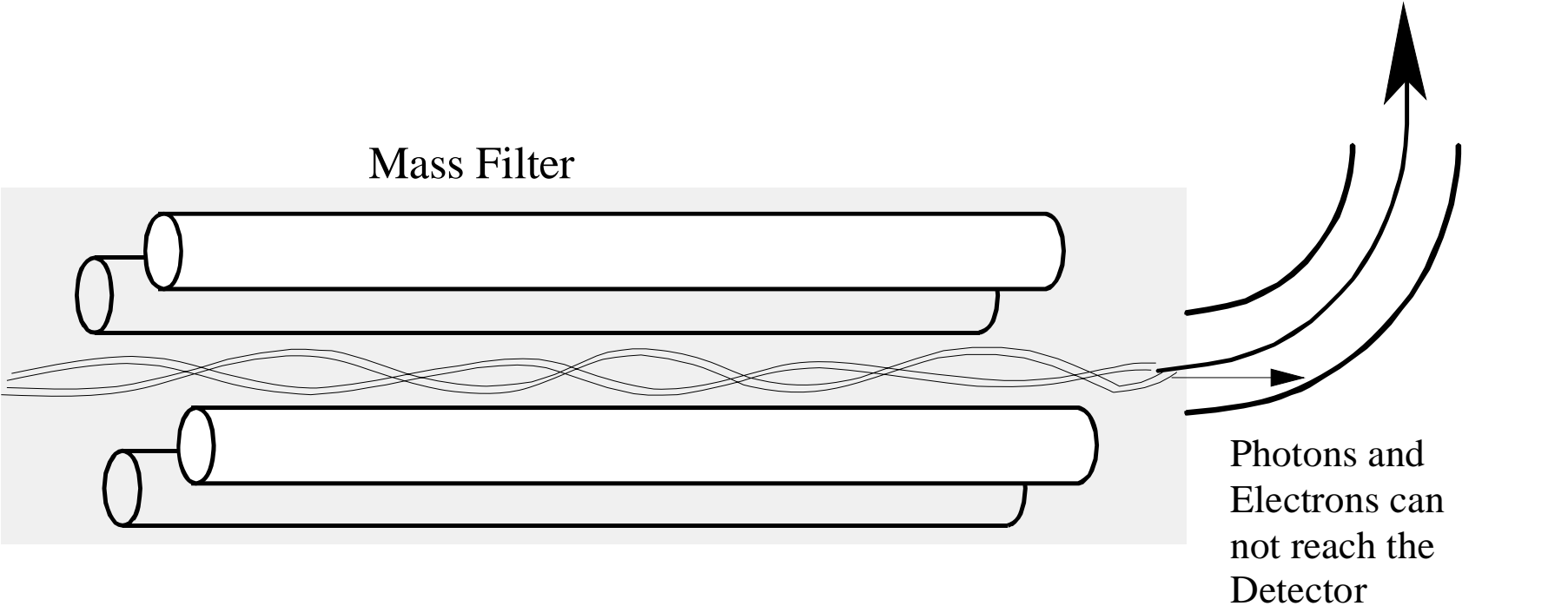


biased system: cathode is most positive electrode



Partial Pressure Gauges

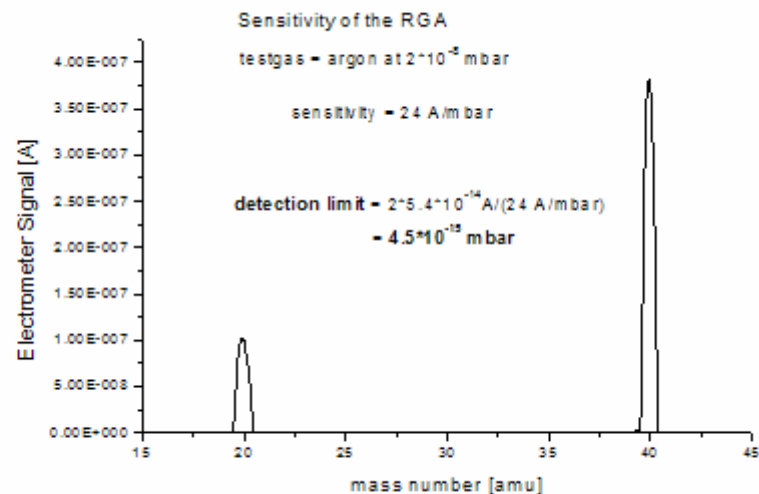
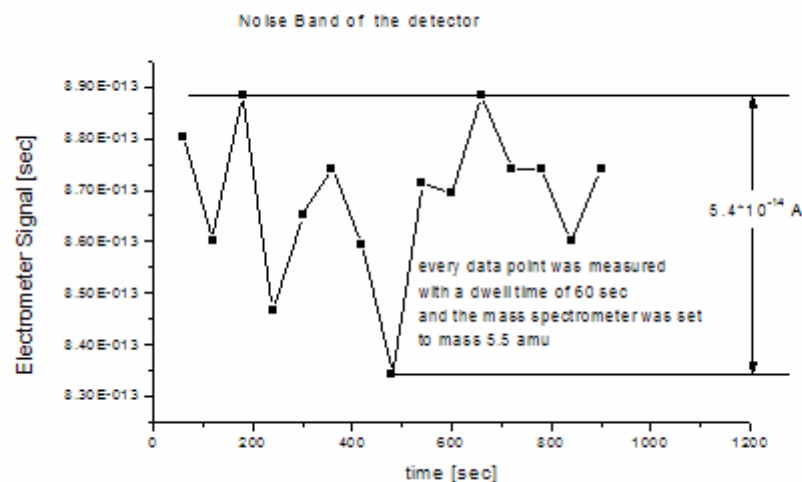
Detection Limit: Reduce Background



Partial Pressure Gauges

Detection Limit

Noise Band and Sensitivity determine the detection limit



Partial Pressure Gauges

Noise band and the sensitivity increase with increasing SEM-Voltage
→ search for an optimum:

1600 VDC Sensitivity: 9.35 A/mbar
Noise band $9.12 \cdot 10^{-14}$ A
Detection Limit $9.75 \cdot 10^{-15}$ mbar

1700 VDC Sensitivity: 24.20 A/mbar
Noise band $1.08 \cdot 10^{-13}$ A
Detection Limit $4.47 \cdot 10^{-15}$ mbar

1750 VDC Sensitivity: 37.90 A/mbar
Noise band $1.56 \cdot 10^{-13}$ A
Detection Limit **$4.13 \cdot 10^{-15}$ mbar**



Partial Pressure Gauges

Requirements to a Partial Pressure Gauge used in UHV/XHV

- **Detection Limit $5 \cdot 10^{-15}$ mbar** ,impurities in the residual gas have to be detected, $5 \cdot 10^{-14}$ mbar is already a compromise for other reasons
- **Gauge has to “survive” the baking of the vacuum system**
- **No influence of the gauge to the vacuum**
- **Gauge has to “survive radiation”**

Partial Pressure Gauges

Choice of the material:

- Metal
- Ceramics
- Thermal expansion coefficients of the partners

→ Metal and Al_2O_3 electrical feed-through

Partial Pressure Gauges

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Partial Pressure Gauges

Choice of the Filament- Material

Rhenium:

rather high vapour pressure

used mainly in High Vacuum

Tungsten:

used in the UHV, has a longer
lifetime than Rhenium and a **lower**
vapour pressure

Yt-Oxide on Iridium

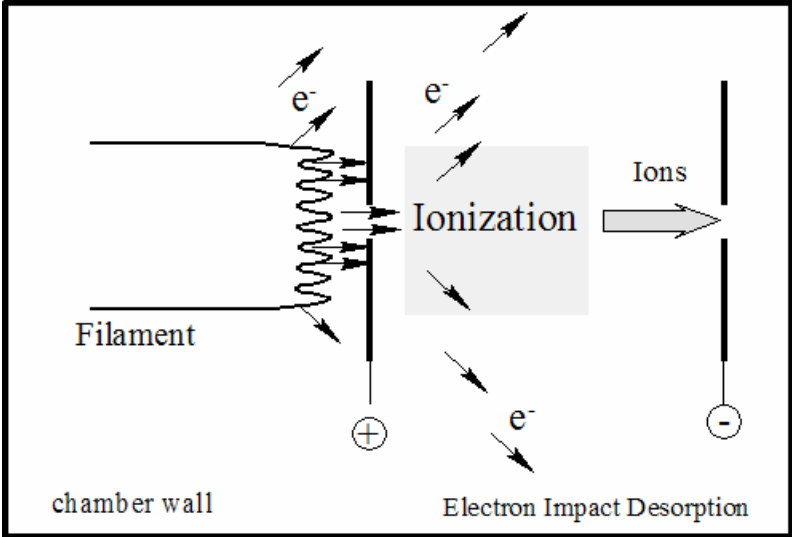
rather resistant against air,
temperature is much lower
compared to pure metal



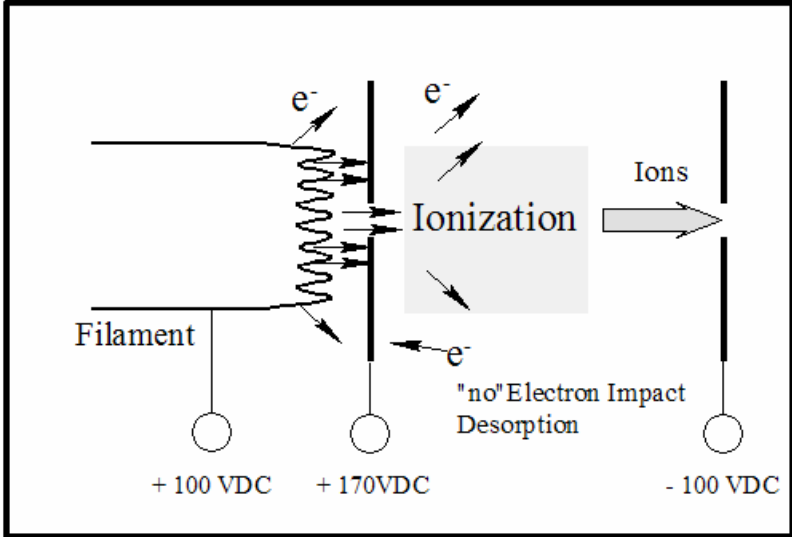
Partial Pressure Gauges

Avoid Electron Induced Desorption

most used technique: no electrical bias



biased system: cathode is most positive electrode



Partial Pressure Gauges

Avoid degassing of the gauge:

- By design no gaps, if threads are necessary they have to have additional holes for degassing,..
- By the choice of material again metal and “real Al_2O_3 ”
- By pre-treatment of the components → vacuum firing at 900°C prior to assembly
- The SEM also has to be consequent UHV-designed
- → then $< 10^{-10}$ mbar l^*s^{-1} degassing is achieved

Partial Pressure Gauges

What about the filament?:

The filament of a gauge is at $> 1000^{\circ}\text{C}$

Therefore the filament has to be switched on during the baking of the vacuum system, otherwise,.....

Partial Pressure Gauges

Requirements to a Partial Pressure Gauge used in UHV/XHV

- **Detection Limit $5 \cdot 10^{-15}$ mbar** ,impurities in the residual gas have to be detected, $5 \cdot 10^{-14}$ mbar is already a compromise for other reasons
- Gauge has to “survive” the baking of the vacuum system
- No influence of the gauge to the vacuum
- Gauge has to “survive radiation”

Partial Pressure Gauges

Separate analyzer and electronics to be able to shield the electronics



Sometimes longer distances than shown here are required. This has an effect on the sensitivity!



Partial Pressure Gauges

Partial Pressure Gauge which can be used in UHV/XHV

- **Detection Limit $5 * 10^{-15}$ mbar , or even below**
- **All metal-ceramic design, according to UHV-design rules**
- **Material is pre-cleaned, vacuum fired for lowest degassing**
- **Gauge and electronics are separated because of radiation**



Partial Pressure Gauges



Tuesday, May 30, 2006

Günter J. Peter

Partial Pressure Gauges

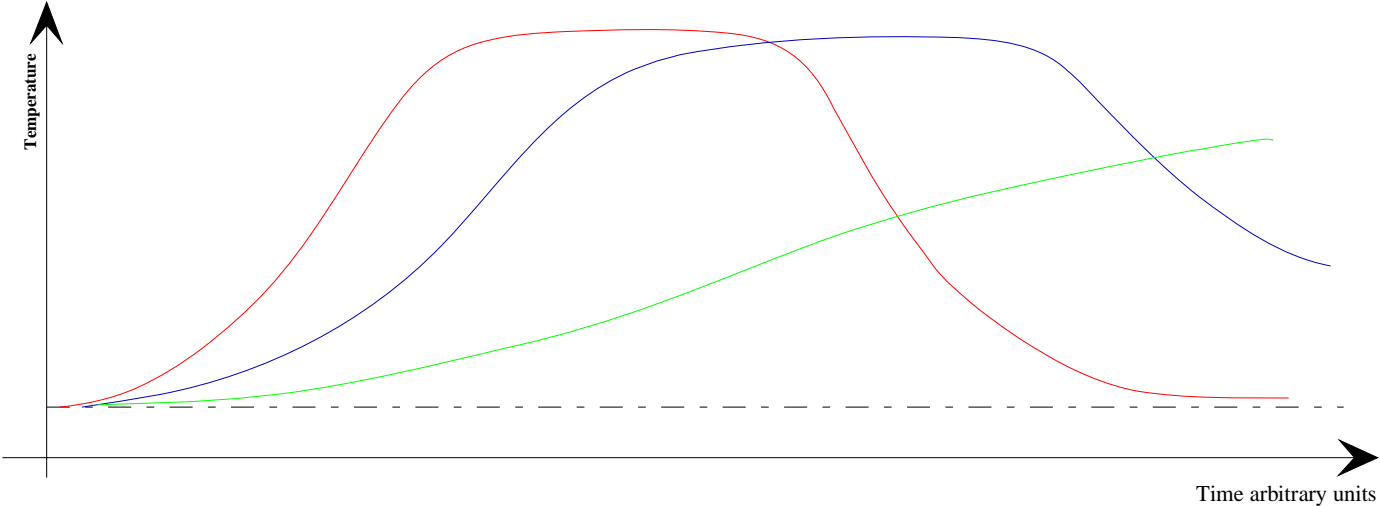
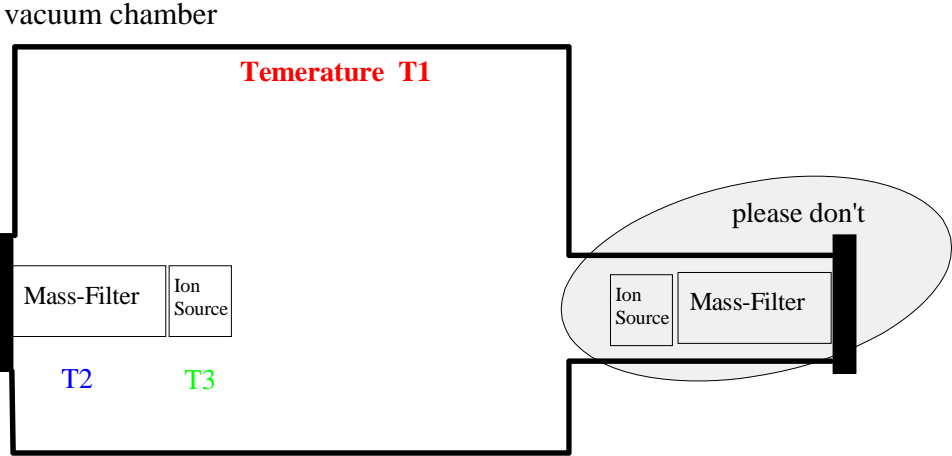
How to operate a partial pressure gauge in the UHV

**few remarks to the
state of the art practise**



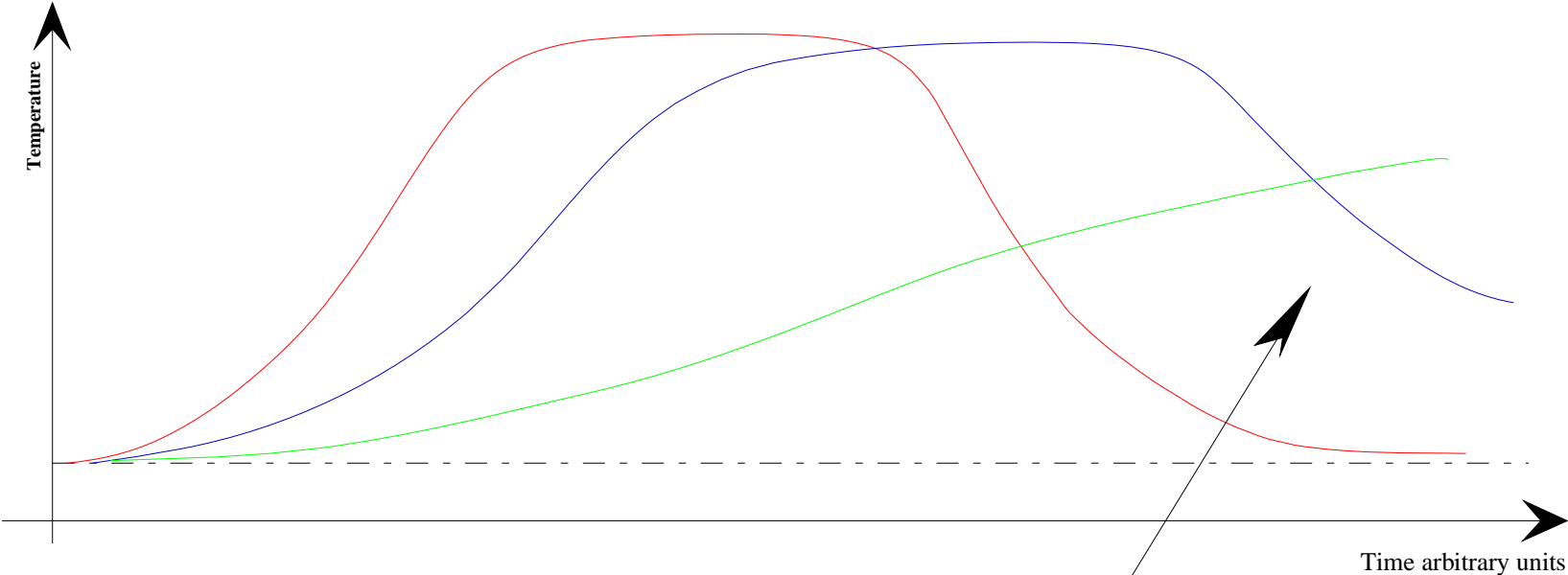
Partial Pressure Gauges

How to bake an RGA:



Partial Pressure Gauges

How to bake an RGA:



Desorption from the walls starts

partly condensation onto the mass filter and here especially

condensation onto the Ion Source



Partial Pressure Gauges

How to bake an RGA:

If possible the instrument has to be installed inside the chamber not inside a tube.

Avoid „cold spots“ at the instrument and inside the chamber wherever possible.
Otherwise there might be condensation onto this surfaces.

Make sure that the whole system is at equal temperature for a sufficient time.

If ever possible switch on the filament of the RGA during baking.

Thus the Ion Source stays at high temperature and no condensation may occur there.



Partial Pressure Gauges

Results obtained with partial pressure gauges in the UHV

after integral baking at 300° over night

filament switched on during the cool down procedure
after degas at 150°C

Residual gas spectra

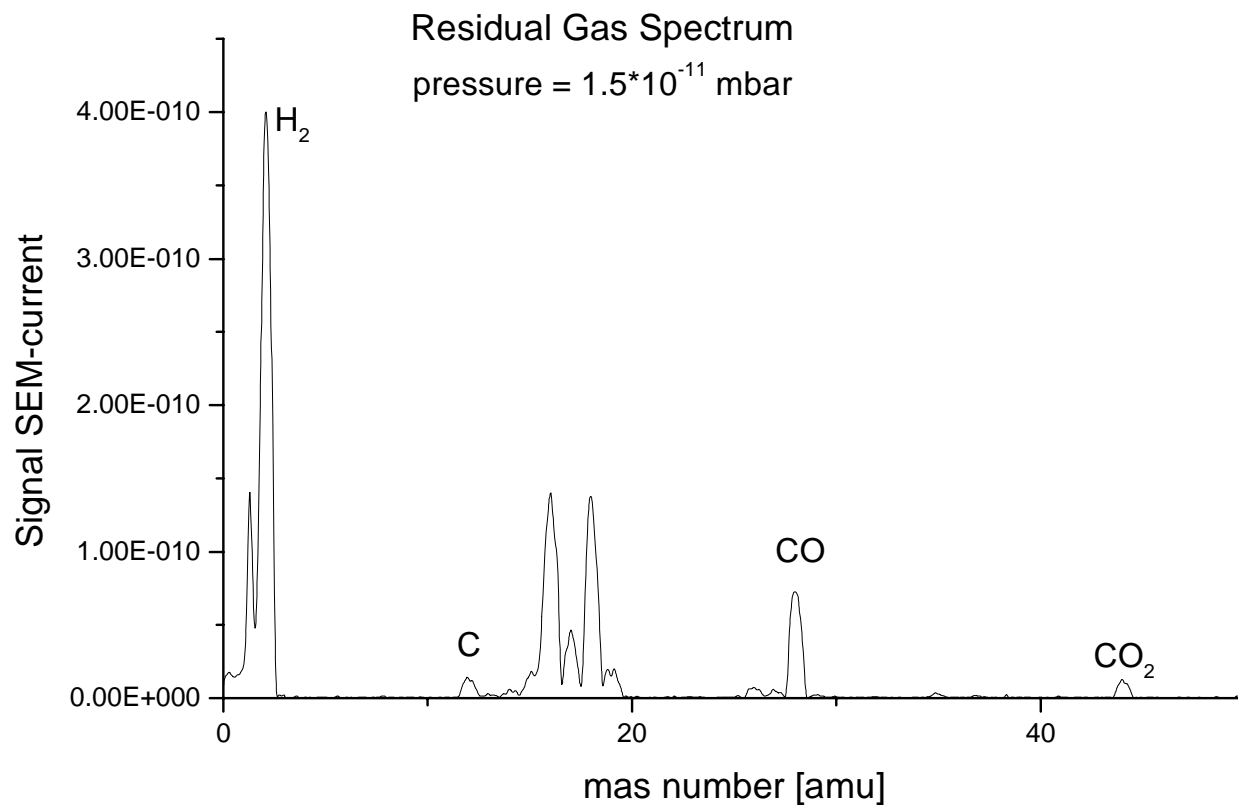
EID-Ions

Leak



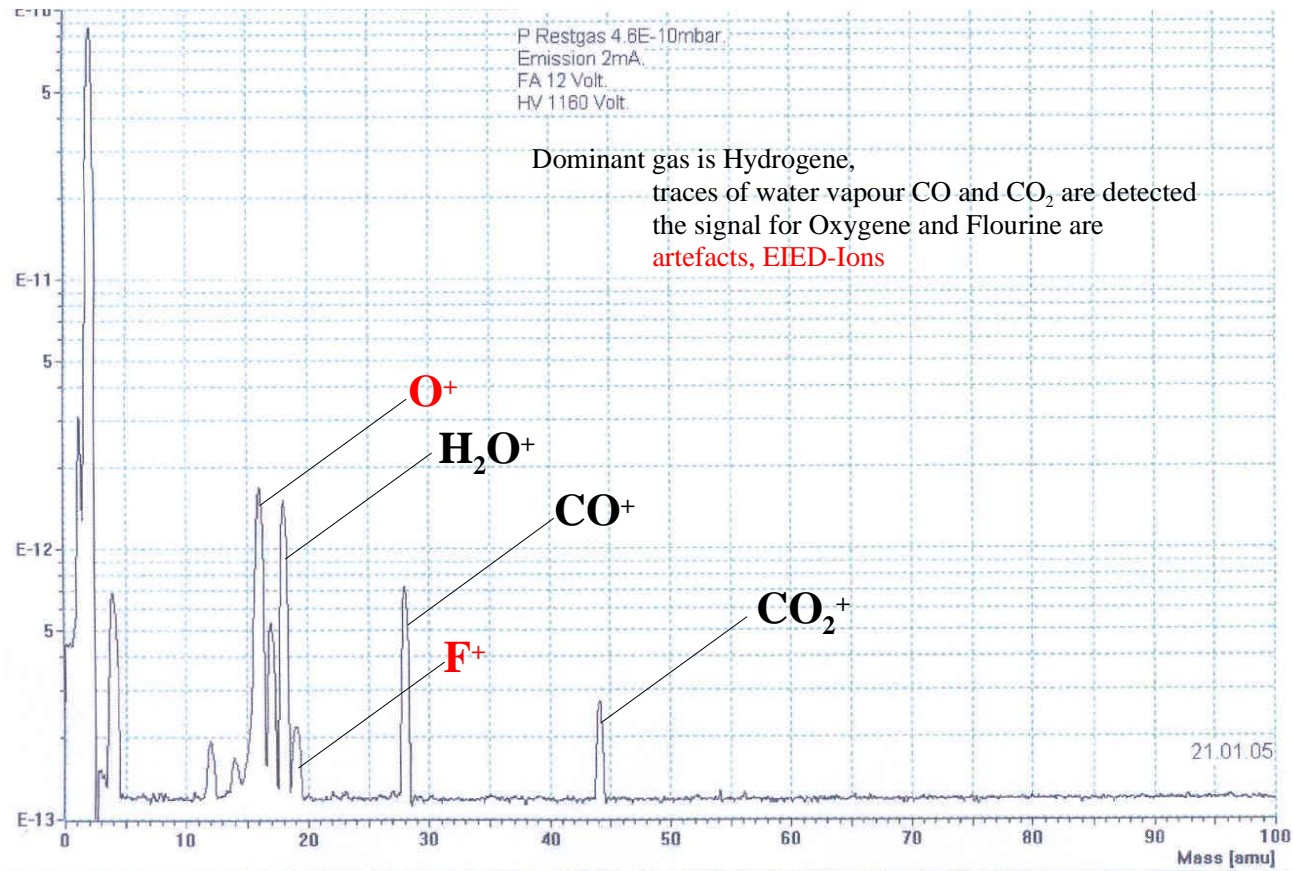
Partial Pressure Gauges

Results achieved



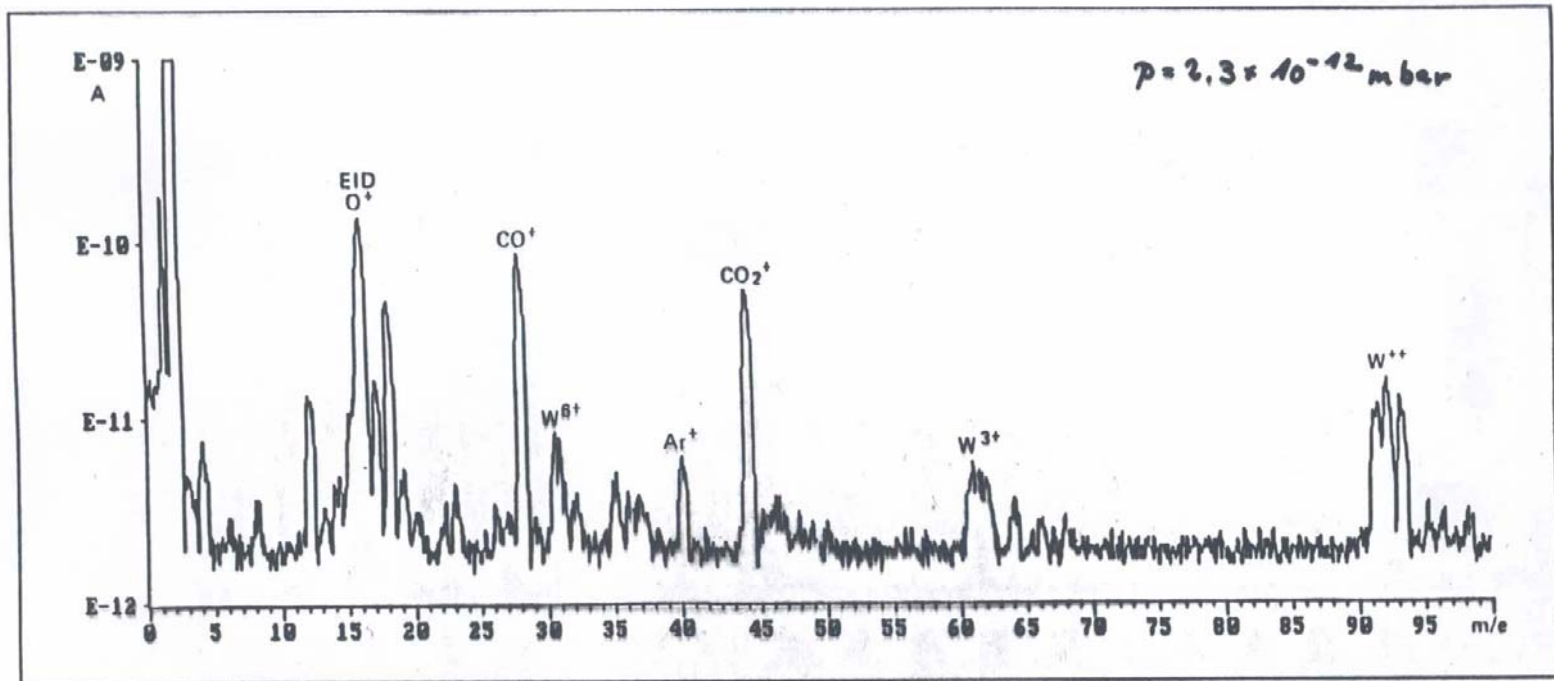
Partial Pressure Gauges

Results achieved : EID-Ions



Partial Pressure Gauges

Results achieved : **EID-Ions**



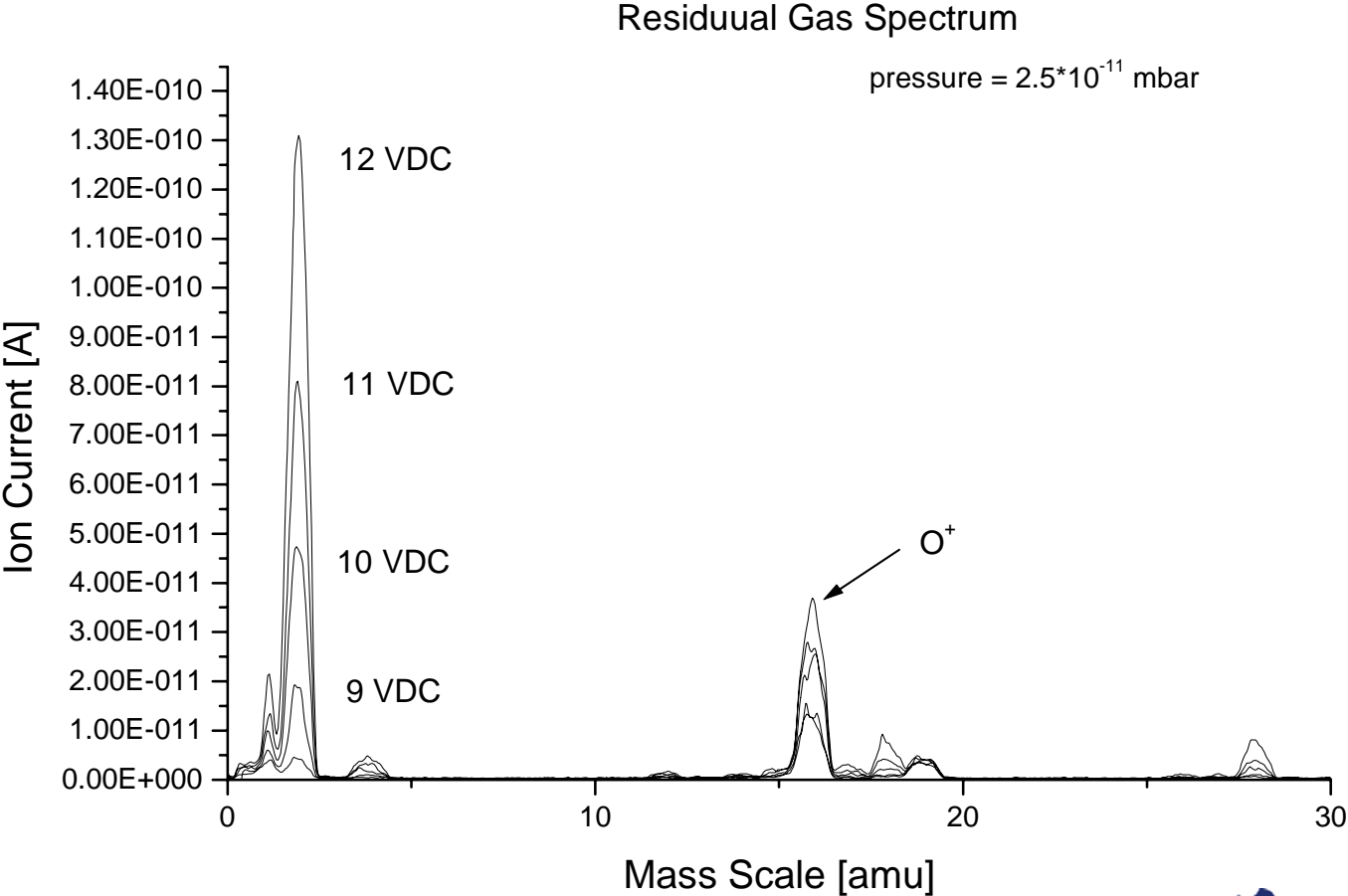
W.K. Huber et al., Vacuum 41, 2103 (1990)

„Physics doesn't change“



Partial Pressure Gauges

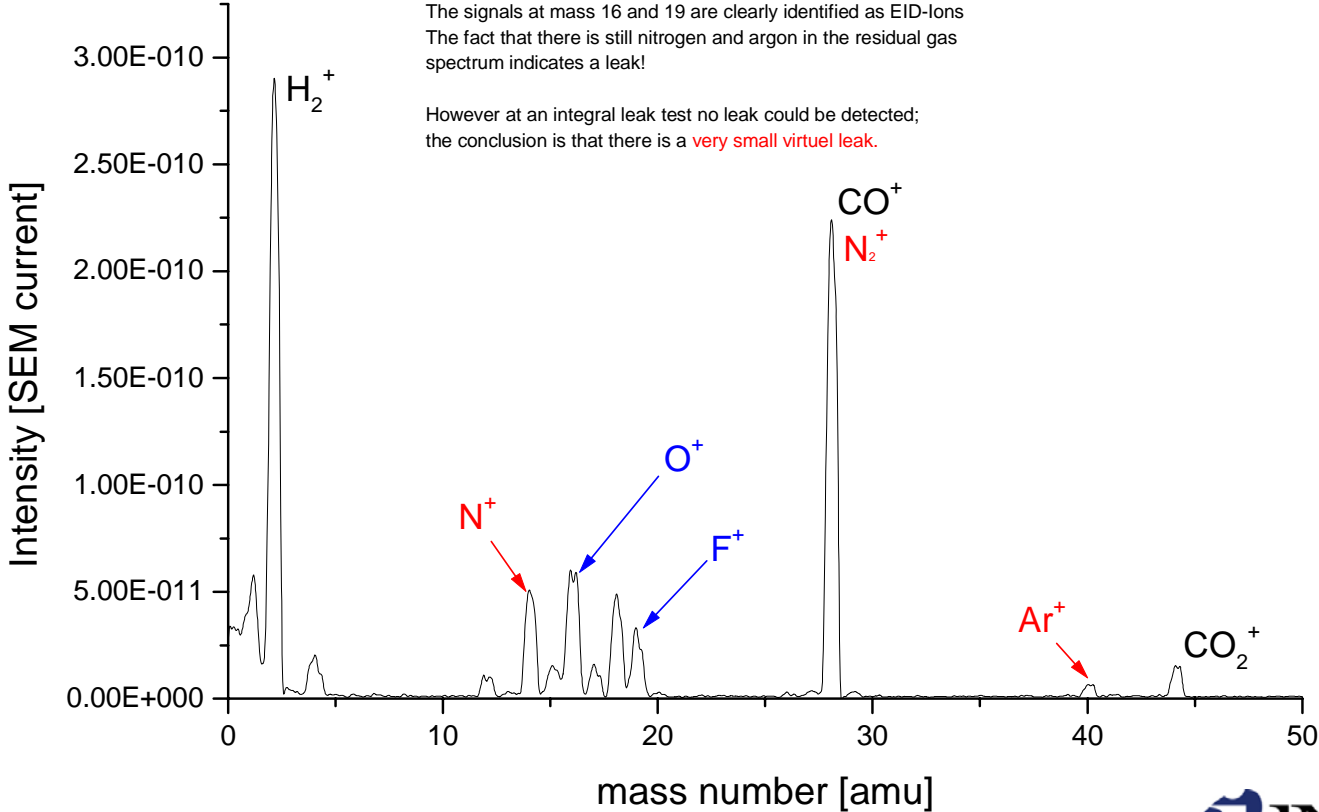
How to identify EID-Ions



Partial Pressure Gauges

Results achieved : **Leak!**

Virtuel Leak at a total pressure of $1.5 \cdot 10^{-11}$ mbar



The signals at mass 16 and 19 are clearly identified as EID-Ions
The fact that there is still nitrogen and argon in the residual gas spectrum indicates a leak!

However at an integral leak test no leak could be detected;
the conclusion is that there is a **very small virtuel leak**.



Partial Pressure Gauges

Results achieved : **Leak!**

- Normally in HV-Systems the “presence of air” in the residual gas, means nitrogen, oxygen and argon, is an indication for a leak.
- Oxygen is a reactive gas which is gettered at the inner surface of an UHV-system. Therefore nitrogen and argon are an indication for a leak in this pressure range; oxygen is not observed.
- In the present case no leak from the outside could be detected during an integral leak test.
- Hypothesis was that there was some gas-bubble inside the vacuum. Later an inclusion at a gasket could be identified.
- After a new gasket was used and after an additional baking the leak “vanished”.



Partial Pressure Gauges

Summary Conclusions:

- Gauges to be used in UHV/XHV require special electrical and mechanical design and pre-cleaned components.
- Because of the very high sensitivity achieved artefacts such as EID-Ions are found in the spectra.
- These instruments are used for trace analysis and in addition
- Can and should be used for leak-detection because of their high sensitivity.

Partial Pressure Gauges

I would like to thank Hans Zogg, who did the most part of the practical work.



Thank you for your attention!

