

## Vacuum Systems

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Vacuum, Surfaces & Coatings Group Technology Department

### Outline

- 1. Vacuum Basis
- 2. Vacuum Components
- 3. Vacuum with Beams : LHC Example



## 1. Vacuum Basis



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### **Units**

• The pressure is the force exerted by a molecule per unit of surface :  $1 Pa = 1 N/m^2$ 

~	Pa	kg/cm <sup>2</sup>	Torr	mbar	bar	atm
1 Pa	1	10.2 10-6	7.5 10-3	10-2	10-5	9.81 10-6
1 kg/cm <sup>2</sup>	98.1 10 <sup>3</sup>	1	735.5	980	0.98	0.96
1 Torr	133	1.35 10-3	1	1.33	1.33 10-3	1.31 10-3
1 mbar	101	1.02 10-3	0.75	1	10-3	0.98 10-3
1 bar	1.01 10 <sup>5</sup>	1.02	750	10 <sup>3</sup>	1	0.98
1 atm	101 300	1.03	760	1 013	1.01	1

### As a consequence of the « vacuum force » ...

Ø (mm)	16	35	63	80	100	130	150	212
kg	2	10	32	52	81	137	182	363



### **Ideal Gas Law**

• Statistical treatment which concerns molecules submitted to thermal agitation (no interaction between molecules, random movement, the pressure is due to molecules hitting the surface)

• For such a gas, the pressure, P [Pa], is defined by the gas density, n [molecules.m<sup>-3</sup>], the temperature of the gas, T [K] and the Boltzman constant k, (1.38  $10^{-23}$  J/K)

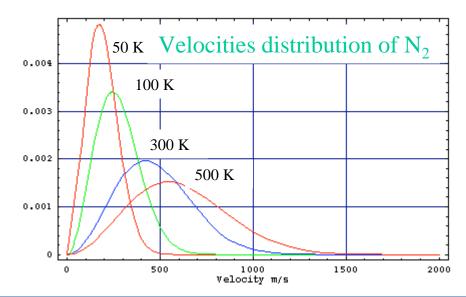
$$P = n k T$$

- The distribution of velocities, dn/dv, follows a Maxwell-Boltzmann function
- The average velocity is :

$$\overline{v} = \sqrt{\frac{8kT}{\pi m}} = 146\sqrt{\frac{T}{M}}$$

• At room temperature (m/s) :

He	Air	Ar
1800	470	400





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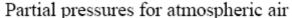
### **Total Pressure and Partial Pressure**

• The gas is usually composed of several types of molecules (ex : air, residual gas in vacuum systems)

• The total pressure, P<sub>Tot</sub>, is the sum of all the partial pressure, P<sub>i</sub> (Dalton law)

$$\mathbf{P}_{\mathrm{Tot}} = \sum \mathbf{P}_{\mathrm{i}} = k \, \mathrm{T} \sum \mathbf{n}_{\mathrm{i}}$$

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	Gas	%	Pi (Pa)
	N <sub>2</sub>	78.1	$7.9 \ 10^4$
	O <sub>2</sub>	20.5	$2.8 \ 10^3$
	Ar	0.93	$1.2 \ 10^2$
	$CO_2$	0.0033	4.4
	Ne	1.8 10 <sup>-3</sup>	2.4 10 <sup>-1</sup>
Traces	He	5.2 10 <sup>-4</sup>	7 10 <sup>-2</sup>





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### **Mean Free Path**

• It is the path length that a molecules traverse between two successive impacts with other molecules. It depends of the pressure, of the temperature and of the molecular diameter.

- It increases linearly with temperature
- For air at room temperature :

$$\lambda_{air}[cm] = \frac{510^{-3}}{P[Torr]}$$

- At atmospheric pressure,  $\lambda = 70$  nm
- At 1 Torr,  $\lambda = 50 \ \mu m$
- At 10<sup>-3</sup> Torr,  $\lambda = 5$  cm
- At 10<sup>-7</sup> Torr,  $\lambda = 500$  m

Increasing mean free path when decreasing pressure

• At 10<sup>-10</sup> Torr,  $\lambda = 500$  km



### **Turbulent and Viscous Flows**

• When pumping down from atmospheric pressure, the physics is caracterised by different flow regimes. It is a function of the pressure, of the mean free path and of the components dimensions.

- Reynold number, Re :
  - if Re > 2000 the flow is turbulent
  - it is viscous if Re < 1000

$$Re = \frac{Q[Torr.l/s]}{0.089D[cm]}$$

• The turbulent flow is established around the atmospheric pressure

• In the low vacuum (10<sup>3</sup>-1 mbar), the flow is viscous. The flow is determined by the interaction between the molecules themselves. The flow is laminar. The mean free path of the molecules is small compared to the diameter of the vacuum chamber

Viscous flow : 
$$\overline{P} D > 0.5 [Torr.cm]$$



### **Transition and Molecular Flows**

• In the medium vacuum (1-10<sup>-3</sup> mbar), the flow is transitional. In every day work, this range is transited quickly when pumping down vacuum chambers. In this regime, the calculation of the conductance is complex. A simple estimation is obtained by adding laminar and molecular conductances.

• In the high vacuum (10<sup>-3</sup> – 10<sup>-7</sup> mbar) and ultra-high vacuum (10<sup>-7</sup>–10<sup>-12</sup> mbar), the flow is molecular. The mean free path is much larger than the vacuum chamber diameter. The molecular interactions do not longer occurs. Molecules interact only with the vacuum chamber walls

Molecular flow :  $\overline{P} D < 1.510^{-2} [Torr.cm]$ 

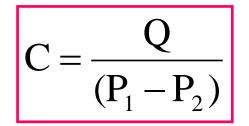
Molecular flow is the main regime of flow to be used in vacuum technology

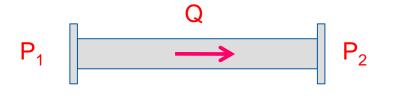
In this regime, the vacuum vessel has been evacuated from its volume. The pressure inside the vessel is dominated by the nature of <u>the surface</u>.



### Conductance

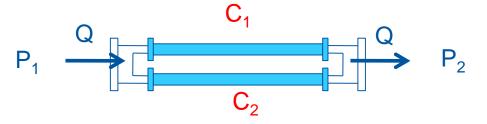
• It is defined by the ratio of the molecular flux, Q, to the pressure drop along a vacuum vessel. It is a function of the shape of the vessel, the nature of the gas and its temperature.



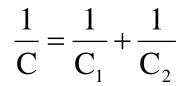


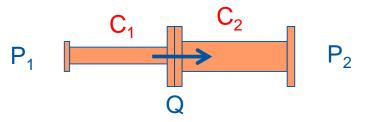
Adding conductances in parallel

$$\mathbf{C} = \mathbf{C}_1 + \mathbf{C}_2$$



Adding conductances in series







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### **Conductance Calculus in Molecular Regime**

•For an orifice :

C = 
$$\sqrt{\frac{kT}{2\pi m}}$$
A; C<sub>air, 20°</sub> [l/s] = 11.6 A[cm<sup>2</sup>]

The conductance of an orifice of 10 cm diameter is 900 l/s

• For a tube :  

$$C = \frac{1}{6} \sqrt{\frac{2\pi kT}{m}} \frac{D^{3}}{L}; \quad C_{air, 20^{\circ}}[l/s] = 12.1 \frac{D[cm]^{3}}{L[cm]}$$

The specific conductance of a tube of 10 cm diameter is 120 l/s.m

To increase the conductance of a vacuum system, it is better to have a vacuum chamber with large diameter and short lenght



### **Pumping Speed**

• The pumping speed, S, is the ratio of the flux of molecules pumped to the pressure

$$I/s \rightarrow S = \frac{Q}{P}$$
 mbar. I/s

- S range from 10 to 20 000 l/s
- Q range from 10<sup>-14</sup> mbar.l/s for metalic tubes to 10<sup>-5</sup> 10<sup>-4</sup> mbar.l/s for plastics

3 orders of magnitude for pumping vs 10 orders of magnitude for outgassing

### **Outgassing MUST be optimised to achieve UHV**





•The outgassing rate, q, of a surface is the number of molecules desorbed from a surface per unit of surface and per unit of time

• It is a function of the surface nature, of its cleanliness, of its temperature and of the pump down time.

• In all vacuum systems, the final pressure is <u>driven</u> by the outgassing rate :  $P_{final} = Q/S = q A / S$ 

Plastic surfaces  $q \sim q_0/\sqrt{t}$ Metallic surfaces  $q \sim q_0/t$ 10-8 (Pa.m/s) ourgassing RATE (Pa m<sup>3</sup>s<sup>1</sup>m<sup>2</sup> ថ្មី ថ្មី ថ្មី Taux de désorption 10-4 CO, x 5 000 10-5 10<sup>10</sup> 2 10-6 Unbaked Al 2 5 102 Temps de mise sous vide (h) 20 40 60 80 100 Lexan MR 400 (polycarbonate) PUMPING TIME (HOURS) genre Plexiglas (polyméthacrylate de méthyle) Ertacétal noir (polyacétal) A.G. Mathewson et al. Téflon (polytétrafluoréthylène) après 15 h d'étuvage à 100 °C J.Vac.Sci. 7(1), Jan/Fev 1989, 77-82 (épaisseur de 10 mm)

### **Good Vacuum Design :**

### Use ONLY metallic surfaces and reduce to ZERO the amount of plastics



## **Cleaning Methods**

• Several means are used in vacuum technology to reduce the outgassing rates

- Chemical cleaning is used to remove gross contamination such as grease, oil, finger prints.
- Example of CERN LHC beam screens :

Degreasing with an alkaline detergent at 50°C in an ultrasonic bath Running tap water rinse Cold demineralised water rinse by immersion

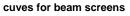
Rinse with alcohol

Dry with ambient air

 Vacuum firing at 950°C is used to reduce the hydrogen content from stainless steel surface

Length: 6 m Diameter: 1 m Maximum charge weight: 1000 Kg Ultimate pressure: 8 10<sup>-8</sup>Torr Pressure at the end of the treatment: high 10<sup>-6</sup> Torr







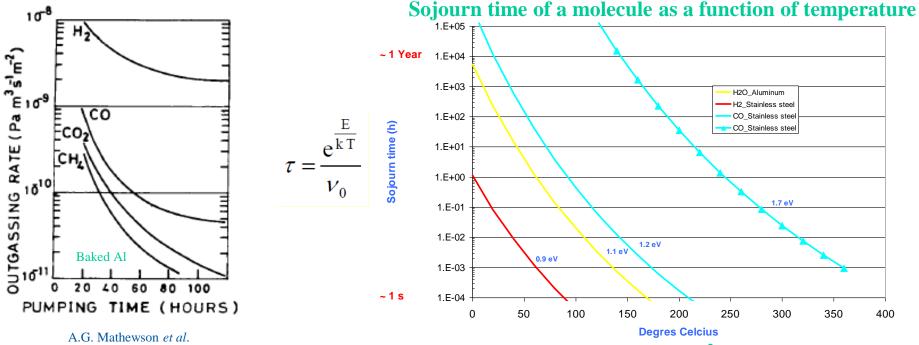
- Glow discharges cleaning is used to remove by sputtering the adsorb gases and the metal atoms
- Wear gloves to handle the material



## In Situ Bake Out

• The outgassing rate of unbaked surfaces is dominated by  $H_20$ .

• A bake-out above 150 degrees increase the desorption rate of  $H_2O$  and reduce the  $H_2O$  sojourn time in such a way that  $H_2$  become the dominant gas



A.G. Mathewson *et al.* J.Vac.Sci. 7(1), Jan/Fev 1989, 77-82

#### Stainless steel after 50 h of pumping (Torr.l/s/cm<sup>2</sup>)

	H2	CH4	H2O	CO	CO2
Unbaked	7 10-12	5 10-13	<b>3 10</b> <sup>-10</sup>	5 10-12	5 10-13
Baked	<b>5 10-13</b>	$5 \ 10^{-15}$	1 10 <sup>-14</sup>	1 10 <sup>-14</sup>	1 10 <sup>-14</sup>



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# 2. Vacuum Components



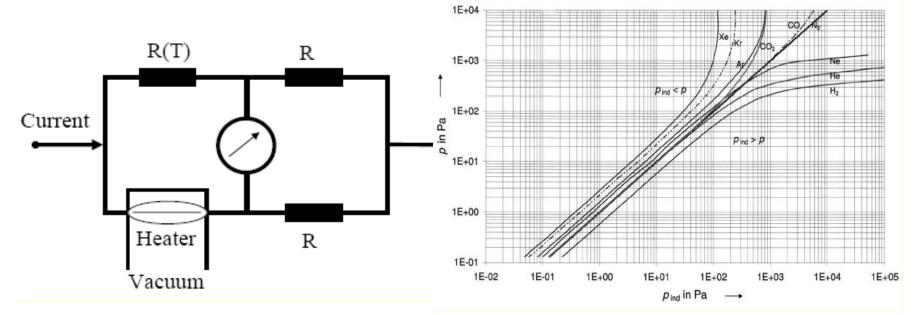
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## **Pirani Gauge**

• Pirani gauges are commonly used in the range 1 atm -10<sup>-4</sup> mbar.

• The operating principle is based on the variation of the thermal conductivity of the gases as a function of pressure. A resistor under vacuum is heated at a constant temperature (~ 120°C). The heating current required to keep the temperature constant is a measure of the pressure.

• In the viscous regime, the thermal conductivity is independent of the pressure. Therefore pressure readings given above 1 mbar are wrong !



#### True vs indicated pressure

K. Jousten. J.Vac.Sci. 26(3), May/Jun 2008, 352-359



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## **Penning Gauge**

•Penning gauges are commonly used in the range 10<sup>-5</sup>-10<sup>-10</sup> mbar. They are use for interlocking purposes

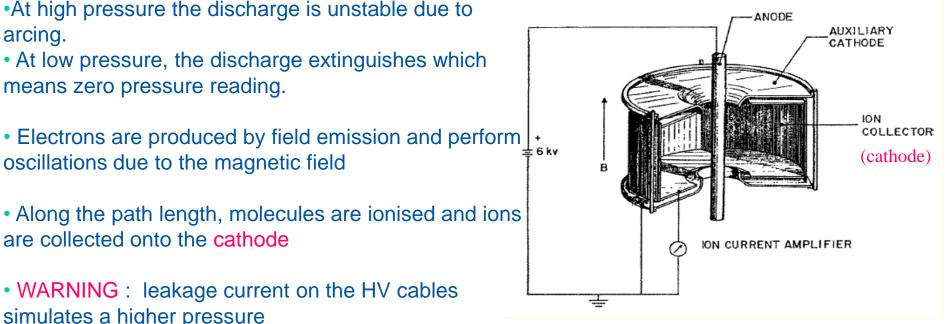
• It is a cold cathode ionisation gauge *i.e.* there are no hot filament

•At high pressure the discharge is unstable due to

• At low pressure, the discharge extinguishes which

• WARNING : leakage current on the HV cables

• The operating principle is based on the measurement of a discharge current in a Penning cell which is a function of pressure :  $I^+ = P^n$ , n is close to 1



P. Redhead. J.Vac.Sci. 21(5), Sept/Oct 2003, S1-S5



arcing.

means zero pressure reading.

are collected onto the cathode

simulates a higher pressure

oscillations due to the magnetic field

### **Bayard-Alpert Gauge**

•Bayard-Alpert gauges are used for vacuum measurement purposes in the range 10<sup>-5</sup> -10<sup>-12</sup> mbar.

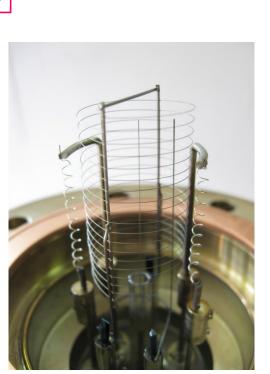
• It is a hot filament ionisation gauge. Electrons emitted by the filament perform oscillations inside the grid and ionise the molecules of the residual gas. Ions are then collected by an electrode.

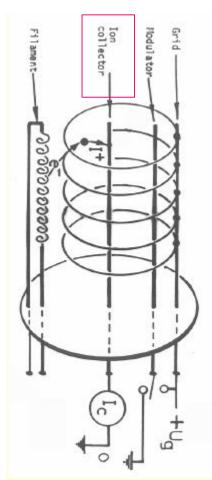


#### Where :

I<sup>+</sup> is the ion current

- I<sup>-</sup> is the filament current
- $\boldsymbol{\sigma}$  is the ionisation cross section
- n the gas density
- L the electron path length
- The gauge needs to be calibrated
- X-ray limit of a ~ 2 10<sup>-12</sup> mbar





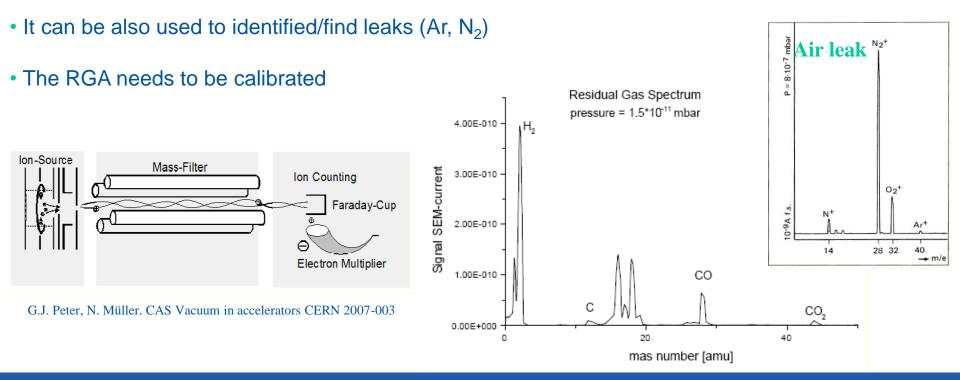


## **Residual Gas Analysers**

• Residual Gas Analysers are used in the range 10<sup>-4</sup> -10<sup>-12</sup> mbar. Their purpose is to do gas analysis

• A filament produces electrons which ionise the residual gas inside a grid. A mass filter is introduced between the grid and the ion collector. The ion current can be measured in Faraday mode or in secondary electron multiplier mode.

• It is a delicate instrument which produces spectrum sometimes difficult to analyse

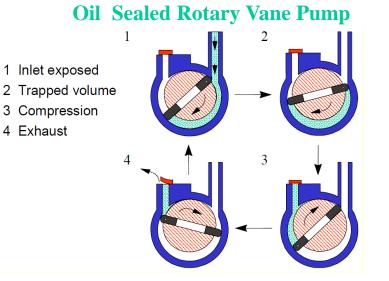


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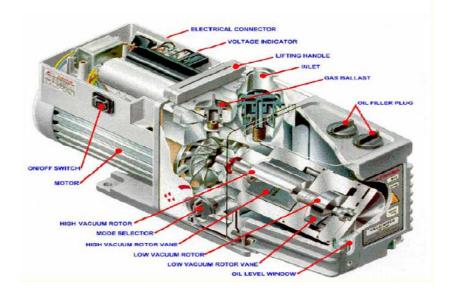
### **Primary Pumps**

•Are used to pump down from atmosphere down to 10<sup>-2</sup> mbar with a speed of a few m<sup>3</sup>/h

- They are usually used as a backing pump of turbomolecular pumps
- Two categories : dry and wet pumps.
- Dry pumps are expensive and need additional cooling (water)
- Wet pumps are operating with oil which acts as a sealing, a lubricant, a heat exchanger and protects parts from rust and corrosion



A.D. Chew. CAS Vacuum in accelerators CERN 2007-003

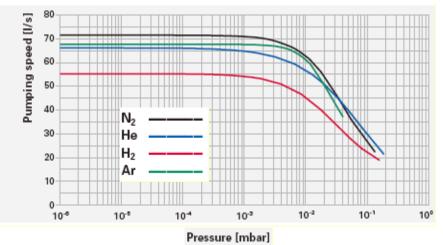




### **Turbomolecular Pump**

• This pump operates in the molecular regime and is used to pump down an accelerator vacuum system. Usually, it is installed with its primary pump on a mobile trolley : it can be removed after valving off

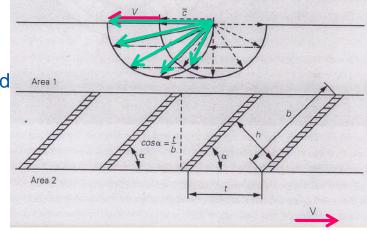
- Its ultimate pressure can be very low : 10<sup>-11</sup> mbar
- Its pumping speed range from 10 to 3 000 l/s



• The pumping mechanism is based on the transfer of impulse. When a molecule collide a blade, it is adsorbed for a certain lenght of time. After re-emission, the blade speed is added to the thermal speed of the molecules. To be significant, the blade speed must be comparable to the thermal speed hence it requires fast moving surfaces (~ 40 000 turns/min)

• The compression ratio ( $P_{inlet}/P_{outlet}$ ) increase exponentially with  $\sqrt{M}$ : "clean" vacuum without hydrocarbons. So, the oil contamination from the primary pump is avoided





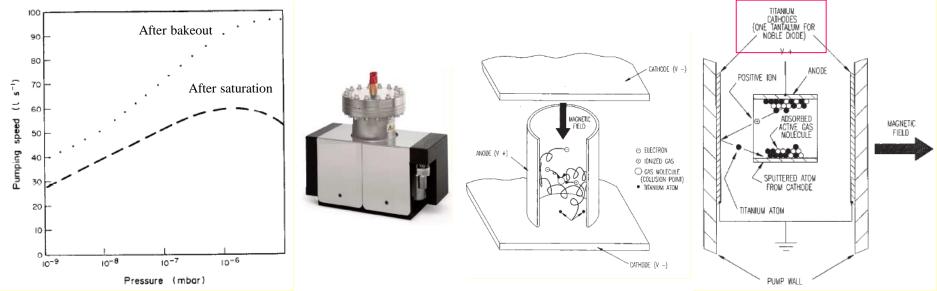


### **Sputter Ion Pump**

•This pump operate in the range 10<sup>-5</sup> -10<sup>-11</sup> mbar. It is used to maintain the pressure in the vacuum chamber of an accelerator.

Their pumping speed range from 1 to 500 l/s

When electrons spiral in the Penning cell, they ionised molecules. Ions are accelerated towards the cathode (few kV) and sputter Ti. Ti, which is deposited onto the surfaces, forms a chemical bounding with molecules from the residual gas. Noble gases and hydrocarbons ,which does not react with Ti, are buried or implanted onto the cathode.
Advantage : like for a Penning gauge, the collected current is proportional to the pressure. It is also used for interlocking.



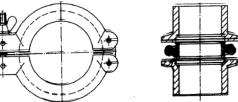
M. Audi. Vacuum 38 (1988) 669-671



### **Flanges and Gaskets**

• For primary vacuum, elastomer seals and clamp flanges are used

 KF type components: Many fittings (elbows, bellows, T, cross, flanges with short pipe, reductions, blank flanges ...) ISO diameters

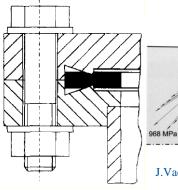


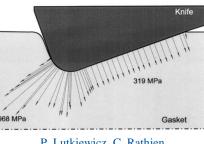


- For ultra high vacuum, metalic gaskets and bolds flanges are used
- Conflat® Type components :

Copper gaskets, blank flanges, rotable flanges, welding flanges, elbows, T, crosses, adaptators, zero length double side flanges, windows ... ISO diameters











### **Tubes, Bellows, Valves**

•Metallic tubes are preferred (low outgassing rate)

 Stainless steel is appreciated for mechanical reason (machining, welding)

• Bellows are equipped with RF fingers (impedance)





Valves are used for roughing and sectorisation

Roughing valve









**Copper tubes** 

**Sector** valves





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## **Leak Detection**

•The vacuum system of an accelerator must be leak tight !

• All vacuum components must follow acceptance tests (leak detection, bake out, residual gas composition and outgassing rate) before installation in the tunnel

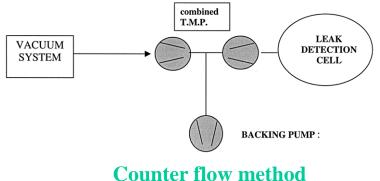
• Virtual leaks, due to a closed volume, must be eliminated during the design phase. Diagnostic can be made with a RGA by measuring the gas composition before and after venting with argon.

#### • Leaks could appear :

during components constructions at welds (cracks or porosity) due to porosity of the material during the assembly and the bake-out of the vacuum system (gaskets)

during beam operation due to thermal heating or corrosion

• Detection method : He is sprayed around the test piece and a helium leak detector (*i.e.* a RGA tune to He signal) is connected to the device under test.



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# 3. Vacuum with Beams : LHC Example



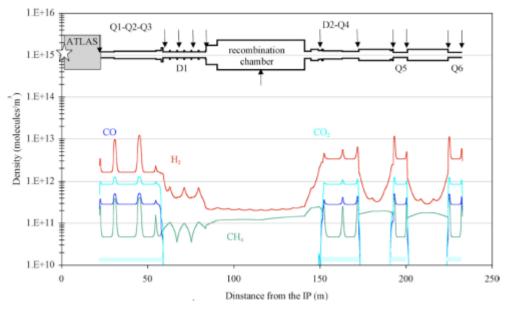
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### **Design value : a challenge with circulating beams**

- Life time limit due to nuclear scattering ~ 100 h
  - n ~ 10<sup>15</sup> H<sub>2</sub>/m3
  - $< P_{arc} > < 10^{-8}$  mbar H<sub>2</sub> equivalent
  - ~ 80 mW/m heat load in the cold mass due to proton scattering

$$\tau = \frac{1}{\sigma c n} \qquad P_{cold mass} = \frac{IE}{c \tau}$$

Minimise background to the LHC experiments



	H2_eq / m3	mbar
<lss<sub>1 or 5&gt;</lss<sub>	~ 5 10 <sup>12</sup>	<b>10</b> <sup>-10</sup>
<atlas></atlas>	~ 10 <sup>11</sup>	<b>10</b> <sup>-11</sup>
<cms></cms>	~ 5 10 <sup>12</sup>	<b>10</b> <sup>-10</sup>

A. Rossi, CERN LHC PR 783, 2004.



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# Why a Challenge?

Because, the static pressure increases by several orders of magnitude due to the dynamics effects related to the presence of a beam

(next 4 slides are just a flavor of the main phenomena which are taking place in an accelerator)



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# 3.1 Dynamic Effects

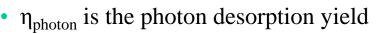


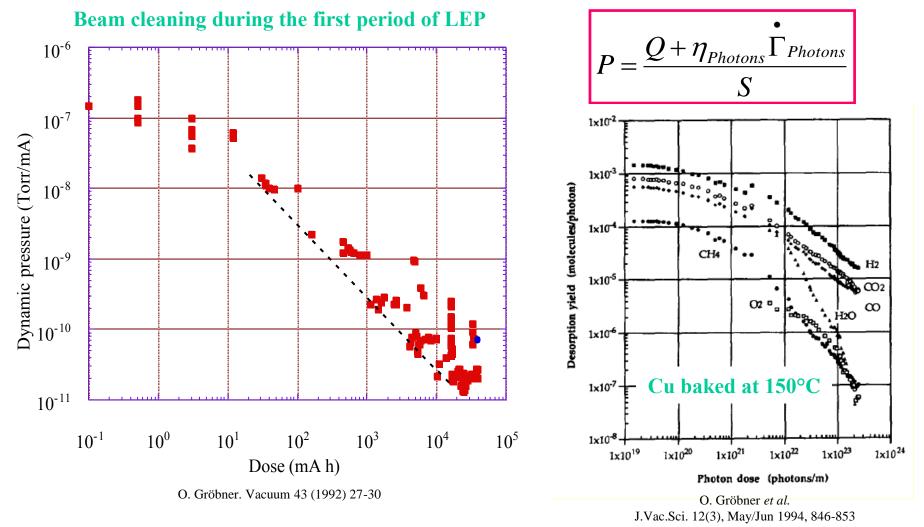
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### **Photon Stimulated Desorption**

•Synchrotron radiation induce gas desorption : SR machine, LEP, LHC

• Heat load and gas load



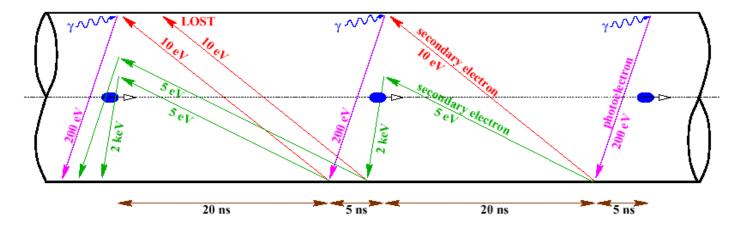




## **Electron Cloud : the Mechanism**

- In modern machine with dense bunches and large positive current : KEK-B, PEP-II, SPS, RHIC, Dafne, LHC, SuperKEKB ...
- Emittance growth, gas desorption and heat load in cryogenic machine
- Key parameters : bunch structure & current vacuum chamber dimension magnetic field secondary electron yield photon electron yield electron and photon reflectivities

$$P = \frac{Q + \eta_{Electrons} \mathring{\Gamma}_{Electrons}}{S}$$



Schematic of electron-cloud build up in the LHC beam pipe.

F. Ruggiero et al., LHC Project Report 188 1998, EPAC 98

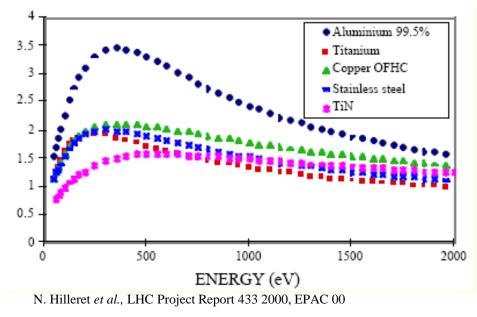


### **Electron Cloud : the Recipes**

•Play with the key parameters :

• ...

- Reduce photoelectron yield (perpendicular vs grazing incidence)
- Reduce secondary electron yields (scrubbing, TiZrV coatings, carbon coatings, geometry ..)
- Reduce the amount of electrons in the system (solenoid magnetic field, clearing electrodes, material reflectivity ...)
- Adapt the bunch structure or the chamber geometry to reduce multiplication



#### **Secondary Electron Yield**



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## **Beam Induced Multipacting along the Beam Pipe**

#### • Key parameters:

- beam structure

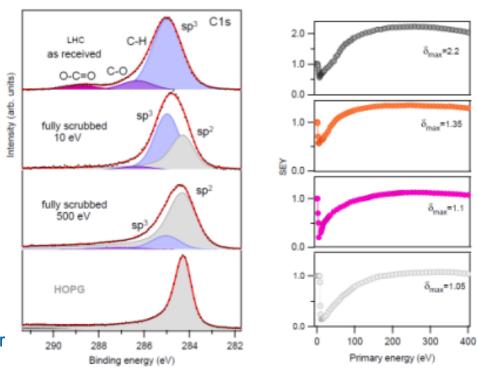
- bunch current

Operational parameters

- vacuum chamber dimension
- secondary electron yield (SEY)
- photoelectron yield
- electron and photon reflectivities
- Mitigations:
  - NEG coating with low SEY (~ 1.1)
  - Beam scrubbing to reduce SEY : Modification of C1s core level Conversion sp<sup>3</sup> => sp<sup>2</sup> High energy electrons increase the number of graphitic like C-C bounds
  - Monitored by ESD reduction

$$P = \frac{Q + \eta_{Electrons}}{S} \overset{\bullet}{\Gamma}_{Electrons}$$





35

R. Cimino et al. PRL 109, 064801(2012)

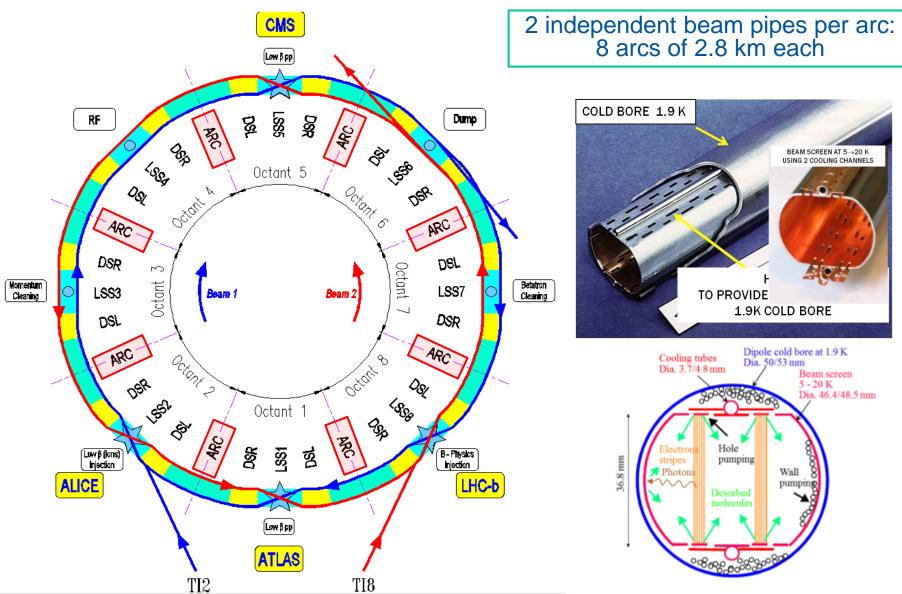
HOPG : highly oriented pyrolitic graphite

## 3.2 Arc Vacuum System



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#### **Cryogenic Beam Vacuum**

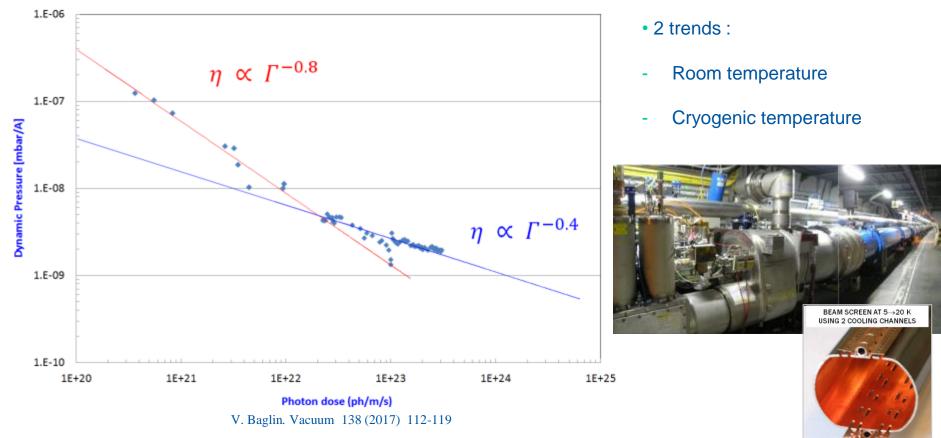


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### **Beam Conditioning under SR**

Arc extremity's vacuum gauges : unbaked Cu and cryogenic beam screen
Reduction by 2 orders of magnitude since October 2010



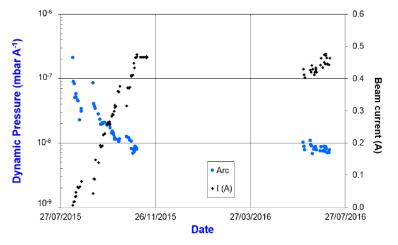
- Inside the arc, at 5-20 K, deltaP < 10<sup>-10</sup> mbar (*i.e.* below detection limit)
- The photodesorption yield at cryogenic temperature is estimated to be < 10<sup>-4</sup> molecules/photon



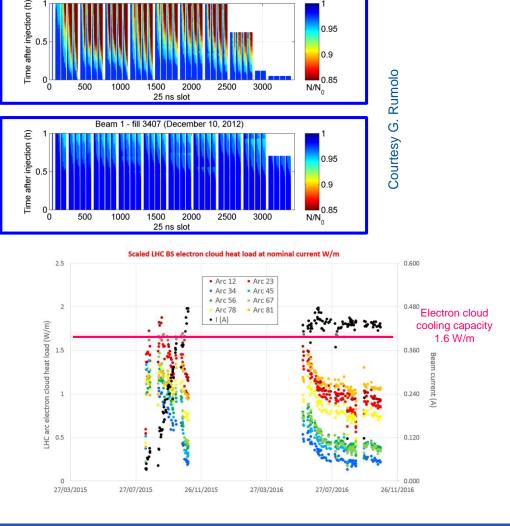
### **Beam Scrubbing**

"Scrubbing" periods are required during LHC commissioning. Particularly during bunch spacing reduction and beam intensity increase

- · Increase of beam life time with time
- Strong pressure reduction in a short time
- · Heat load reduction with time



V. Baglin. Vacuum 138 (2017) 112-119





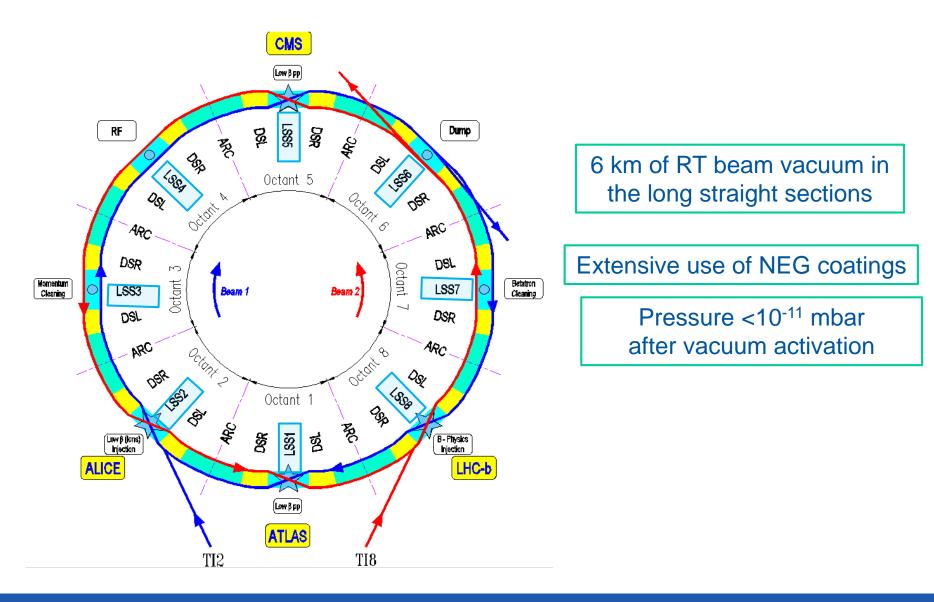
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# 3.3 RT Vacuum System



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#### **Room Temperature Beam Vacuum**



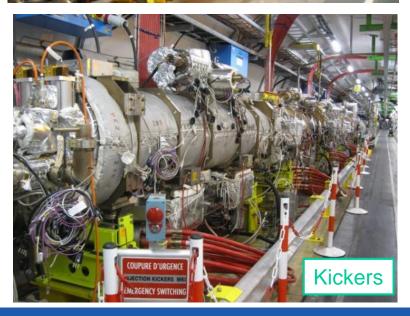


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#### **Standard Components Installed Inside LSS**

• Warm magnets, kickers, septum, collimators, beam instrumentation ...









**Beam Instrumentations** 

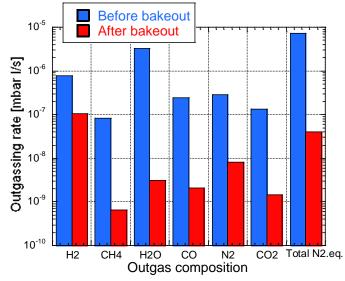


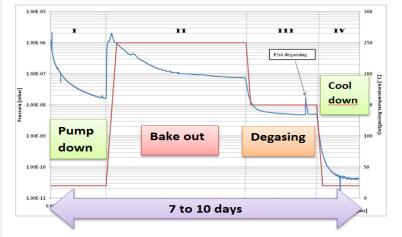
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#### **Vacuum Acceptance Tests**

• Prior installation more than 2300 LSS's equipments have been baked and validated at the surface :

- leak detection
- residual gas composition
- total outgassing rate
- Example : studies for LHC collimators
  - outgassing rate
  - impact on getter coated vacuum chambers





G. Cattenoz et al. IPAC'14, Dresden 2014

Status	Q (mbar I /s)
Unbaked	7 10 <sup>-6</sup>
1st bake-out	7 10 <sup>-8</sup>
2 <sup>nd</sup> bake-out	5 10 <sup>-8</sup>
3rd bake-out	4 10 <sup>-8</sup>

J. Kamiya et al. Vacuum 85 (2011) 1178-1181



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#### **Room Temperature Vacuum System**

- ~ 1  $\mu$ m thick, Non Evaporable Getter TiZrV coated vacuum chambers ensure the required vacuum performances for LHC
- Some vacuum chambers were constructed and getter coated ...



Courtesy R.Veness and P. Chiggiato TE-VSC

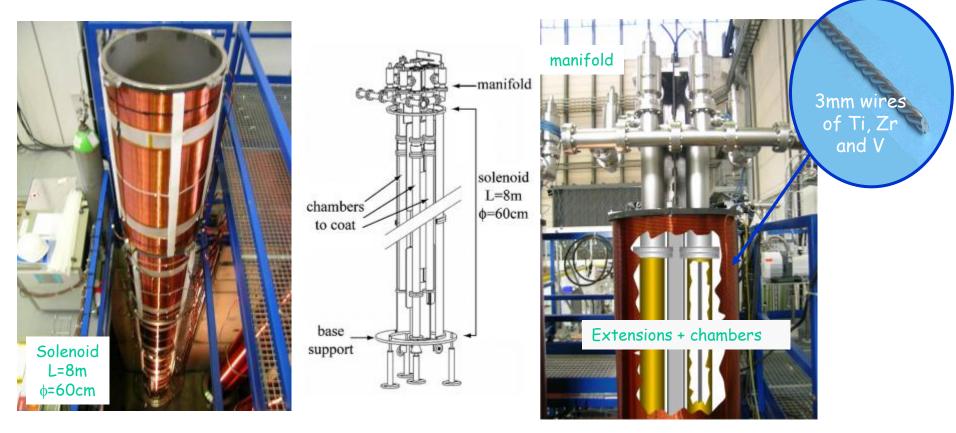


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### **LSS Coating System**

- Ti-Zr-V is coated by magnetron sputtering with Kr gas
- ~ 1  $\mu m$  thick

 All room temperature vacuum chamber including the experimental beam pipe are coated with Ti-Zr-V



P. Costa Pinto, P. Chiggiato / Thin Solid Films 515 (2006) 382-388

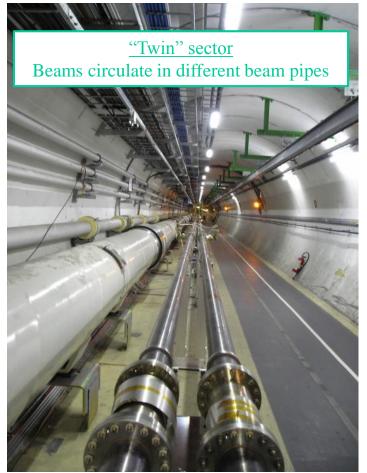


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#### **Room Temperature Vacuum System**

•..... and installed inside the LHC tunnel

• to bring the separated beams from the arcs into a single beam pipe for the experiments (held at room temperature !)



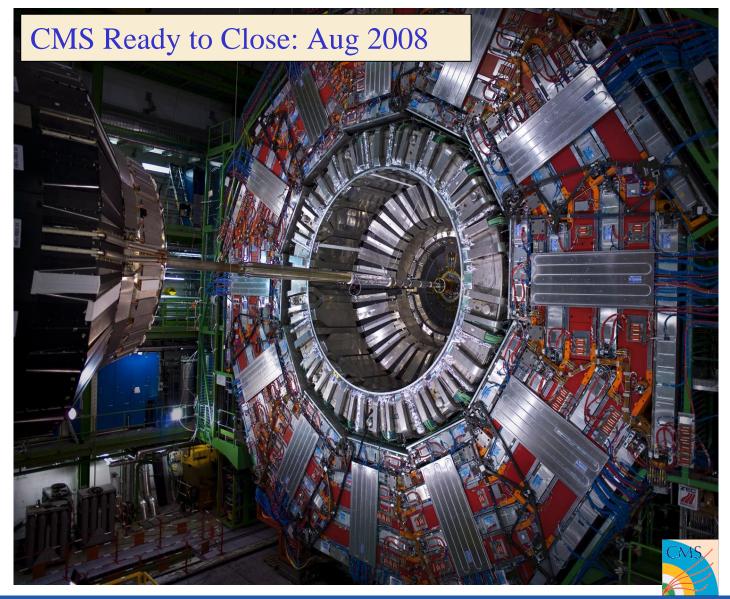


<u>"Combined" sector</u> Both beams circulates in the same beam pipe



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#### And of Course ... Through the LHC Experiments





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#### **Beam Pipe Installation in ATLAS Before Closure**

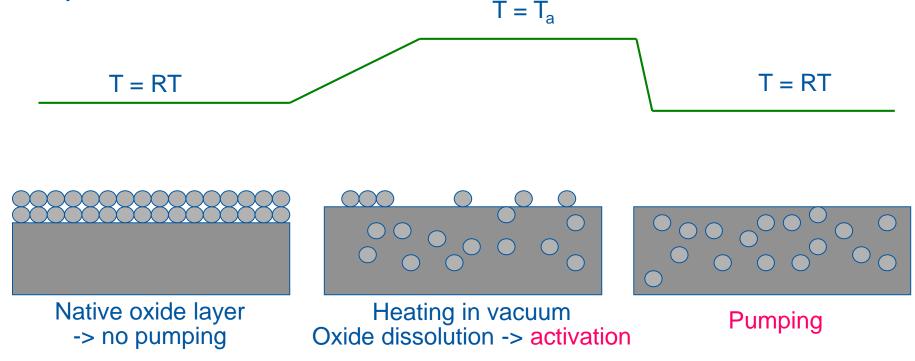




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#### **Non-Evaporable Getter (NEG)**

 Getters are materials capable of chemically adsorbing gas molecules. To do so their surface must be clean. For <u>Non-Evaporable Getters a clean surface is</u> obtained by heating to a temperature high enough to dissolve the native oxide layer into the bulk.



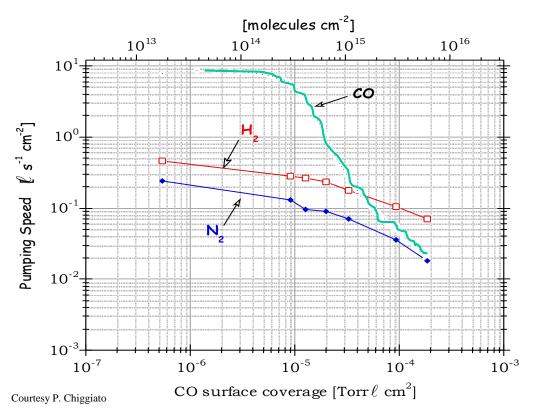
• NEGs pump most of the gas except rare gases and methane at room temperature

P. Chiggiato and P. Costa Pinto, Thin Solid Films, 515 (2006) 382-388



#### **TiZrV Vacuum Performances**

#### **Pumping Speed**



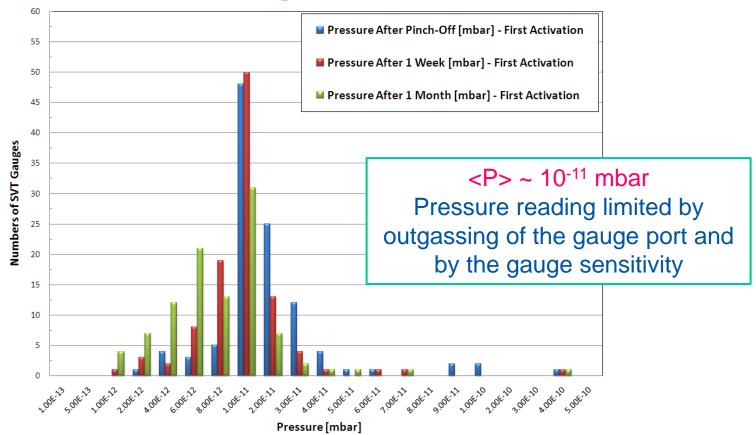
- Very large pumping speed : ~ 250 l/s/m for  $H_2$ , 20 000 l/s.m for CO
- Very low outgassing rate
- But : limited capacity and fragile coating sensitive to pollutant (hydrocarbons, Fluor ...)



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#### Room Temperature Vacuum System : Static Pressure < 10<sup>-11</sup> mbar

Ultimate Vacuum Pressure Distribution after NEG Activation of the LHC Room Temperature Vacuum Sectors



G. Bregliozzi et al. EPAC'08, Genoa 2008



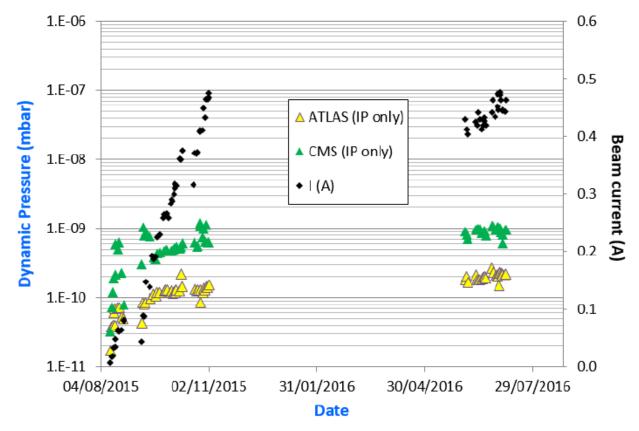
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### **LHC Experimental Areas**

NEG coated vacuum system

=> Large pumping speeds, low SEY and desorption yields

• <P<sub>LHC Experiments</sub> > ~ 5 10<sup>-10</sup> mbar => with 25 ns bunch spacing and 450 mA => No background issues: within specifications



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# 3.4 What about the future? HL-LHC

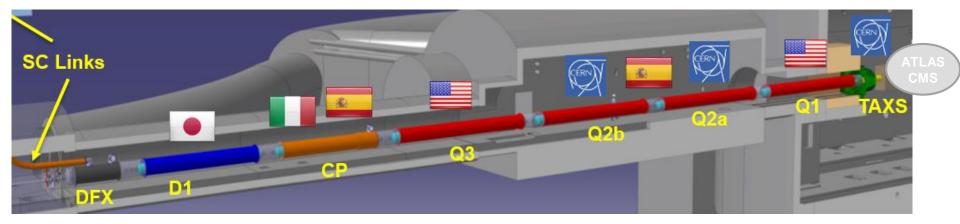




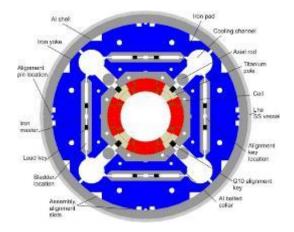
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### NEW focussing quadrupole and merging dipole

• Decrease beta (*i.e* beam size) at collision point (beta\*) from 55 cm to 15 cm



- All superconducting magnets at 1.9 K with a beam screen at 5-20 K or 60-80 K
- •Q1, Q2, Q3, CP (corrector package)
  - Nb<sub>3</sub>Sn (new technology)
  - 150 mm ID, gradient = 130 T/m, peak field 11.5 T
- D1, D2
  - NbTi (classical technology)
  - 150 mm, 5.6 T

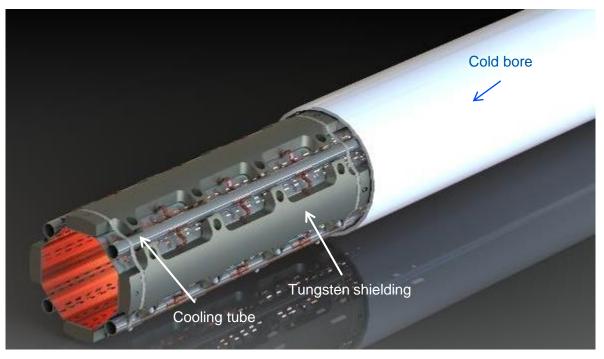




#### **Shielded Triplet Beam Screens**

• Triplets beam screens are shielded with tungsten to intercept the debris produced at the interaction point, protecting thus the cold mass

- Nominal heat load on the beam screen = 15 W/m
- Four cooling tubes extract the beam induced heating and maintain the beam screen temperature along the Triplet string in the 40-60 K temperature range
- Carbon coated beam screen wall to mitigate electron multipacting





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#### **Some References**

- Cern Accelerator School, Vacuum technology, CERN 99-05
- Cern Accelerator School, Vacuum in accelerators, CERN 2007-03
- Cern Accelerator School, Vacuum for particle accelerators, 6-16/2017, Sweden
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- Scientific foundations of vacuum technique, S. Dushman, J.M Lafferty. J. Wiley & sons. Elsevier Science.
- Les calculs de la technique du vide, J. Delafosse, G. Mongodin, G.A. Boutry. Le vide.
- Vacuum Technology, A. Roth. Elsevier Science
   Some Journals Related to Vacuum Technolgy
- Journal of vacuum science and technology
- Vacuum



#### Thank you for your attention !!!



## Spare slides

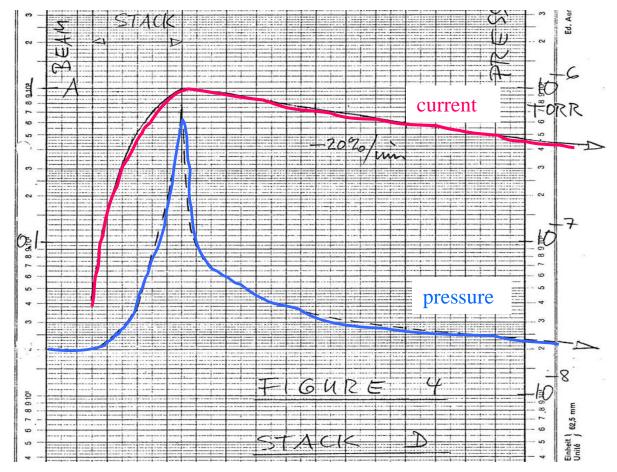


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#### **Vacuum Instability : the Effect**

• In circular machine with large proton current : ISR, LHC

- Beam current stacking to 1 A
- Pressure increases to 10<sup>-6</sup> Torr (x 50 in a minute)
- Beam losses



#### First documented pressure bump in the ISR

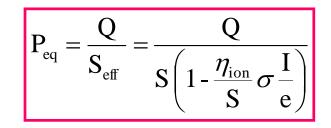
E. Fischer/O. Gröbner/E. Jones 18/11/1970

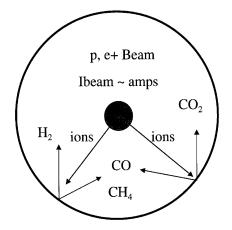


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#### Vacuum Instability : Mechanism and Recipe

- Origin is ions produced by beam ionisation
- Reduction of the effective pumping speed, S<sub>eff</sub>





• When the beam current approach the **critical current**, the pressure increases to infinity

cipe:  
Reduce 
$$\eta_{ion}$$
 Increase pumping speed  
 $Increase pumping speed$ 

A.G. Mathewson, CERN ISR-VA/76-5



• Recipe:

Reduce  $\eta_{ion}$ 

### LHC Beam Screen Stability

• A minimum pumping speed is provided thanks to the beam screen's holes

$$(\eta_i I)_{\text{crit}} = \frac{e}{\sigma} S_{\text{eff}}$$

	H <sub>2</sub>	CH <sub>4</sub>	CO	CO <sub>2</sub>
( <i>ηI</i> ) <sub>crit</sub> [A]	1300	80	70	35

Beam screen's holes provide room for LHC upgrades .....



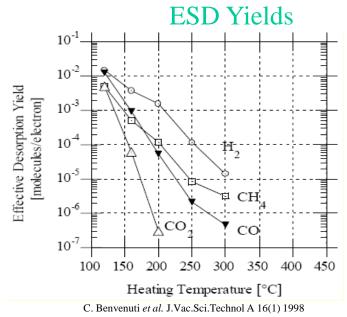
Courtesy N. Kos CERN TE/VSC

• NB : In the long straight sections, vacuum stability is provided by TiZrV films and ion pumps which are less than 28 m apart

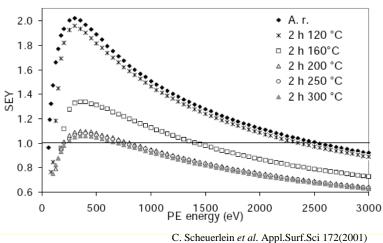


#### **TiZrV Vacuum Performances**

- Very low stimulated desorption yield
- SEY ~ 1.1 => very low multipacting
- But : limited capacity and fragile coating sensitive to pollutant (hydrocarbons, Fluor ...)



#### Secondary Electron Yield



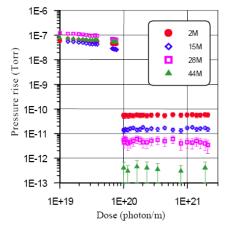


Figure 2: Pressure rise measured in the centre of the TiZrV coated test chamber before activation ( $<1\cdot10^{20}$  photons/m) and after activation ( $>1\cdot10^{20}$  photons/m).

#### **PSD** Yields

Table 2: Summary of results from the activated test chamber				
Gas	Sticking probability	Photodesorption yield (molecules/photon)		
H <sub>2</sub>	~0.007	~1.5.10-5		
$\mathrm{CH}_4$	0	2.10-7		
CO (28)	0.5	<1.10-5		
$C_xH_y(28)$	0	<3.10-8		
CO <sub>2</sub>	0.5	<2.10-6		

V. Anashin et al. EPAC 2002



